



EMORY  
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# Recent Applications of Orthogonality Constrained Density Functional Theory: Core Excitations

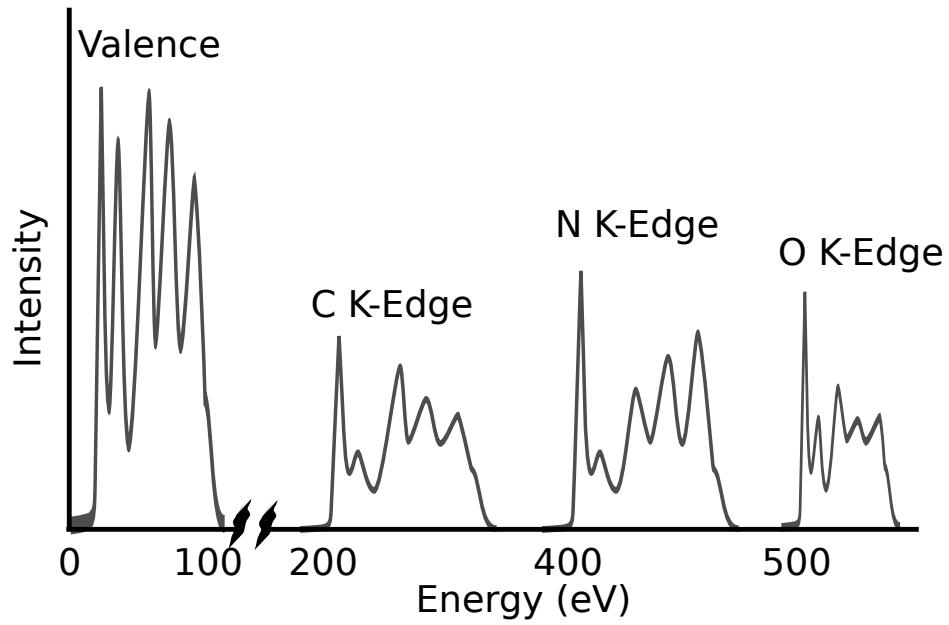
Wallace Derricotte, Prakash Verma, Francesco Evangelista

SETCA Presentation

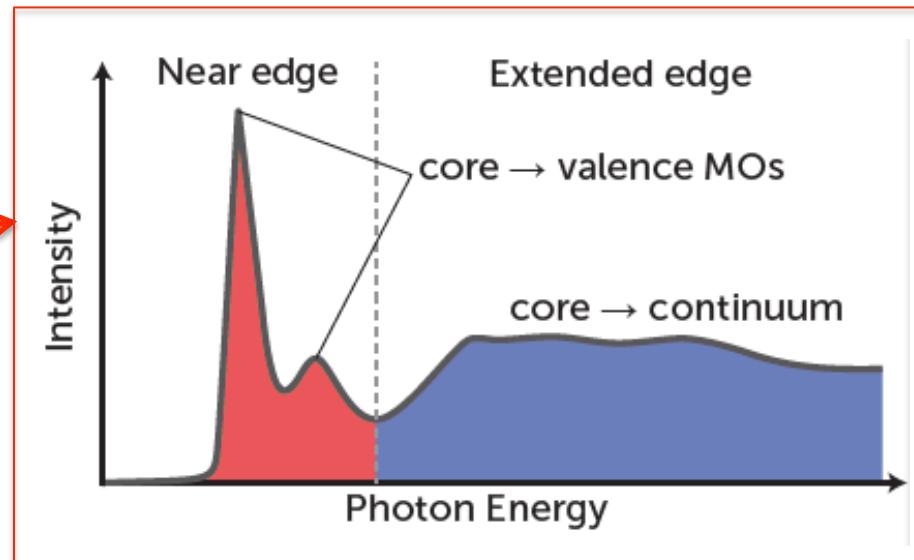
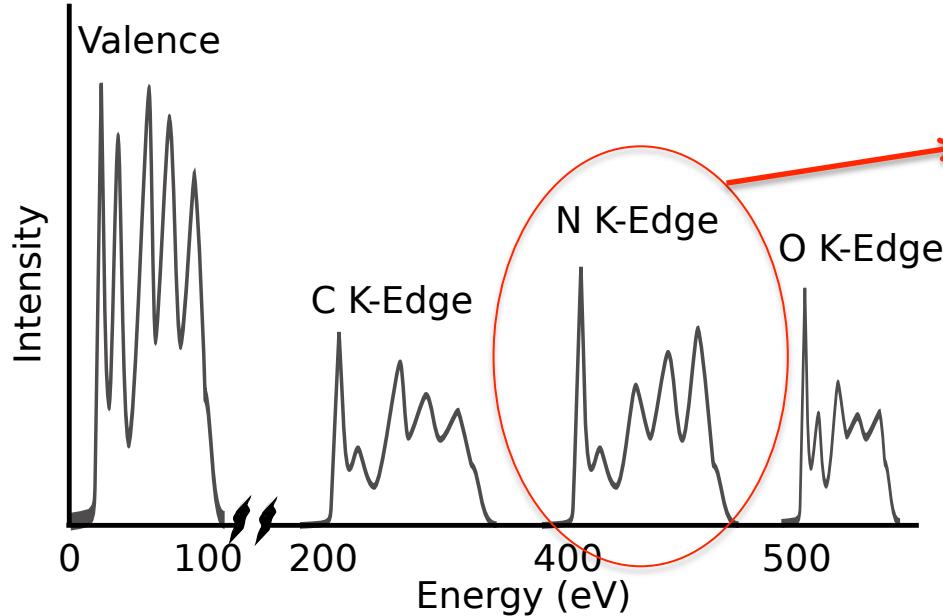
Evangelista Lab

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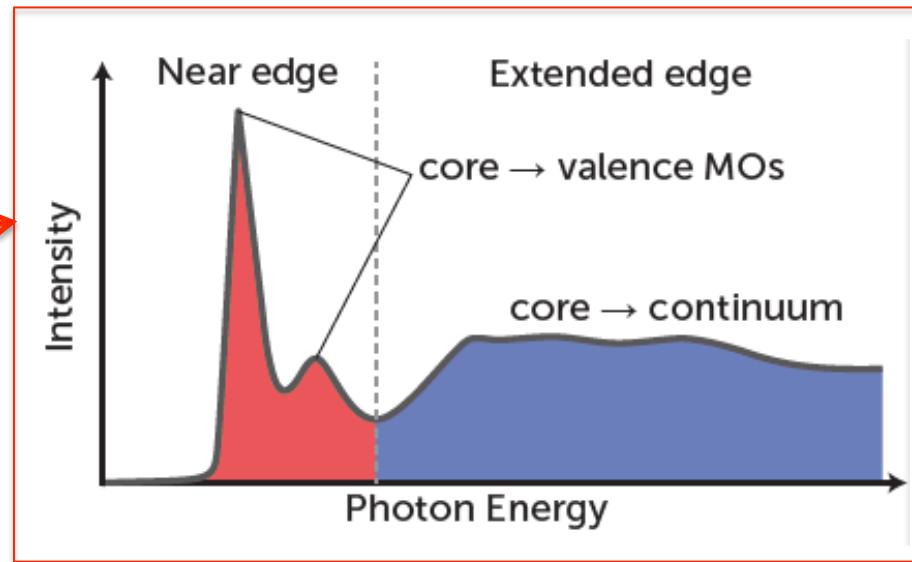
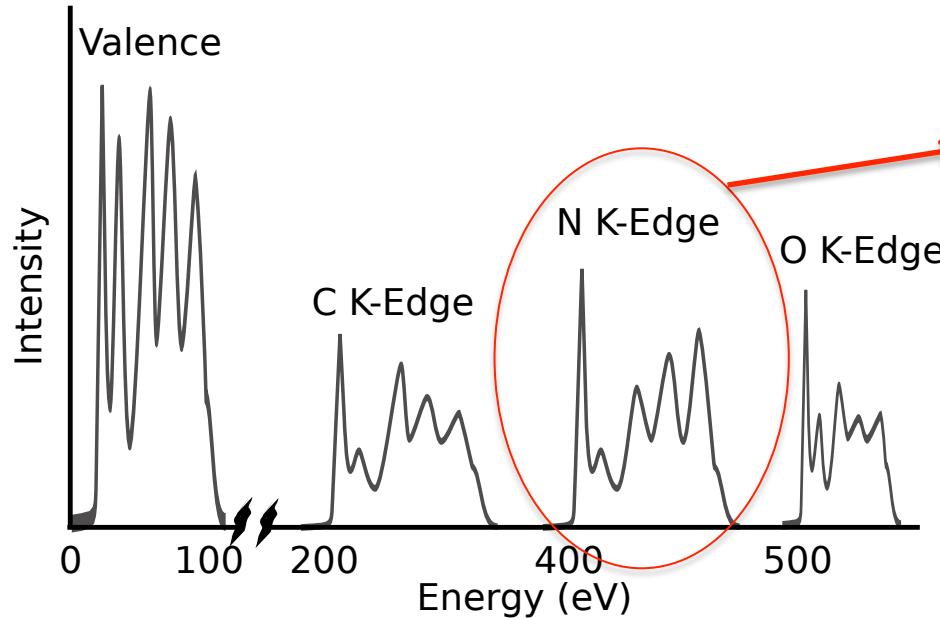
# Core-Excited States



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- Theoretical Challenges Include:
  - Orbital relaxation
  - Require treatment of scalar (1s excitations) and spin-dependent relativistic effects
  - Electron correlation

# Computational Approaches

## Wavefunction/Green's Function Methods

- SOS-CIS(D)<sup>1</sup>
- RASSCF<sup>2</sup>
- LR-CC methods<sup>3</sup>
- Algebraic Diagrammatic Construction (ADC)<sup>4</sup>

## TDDFT

- Linear Response TDDFT<sup>5</sup>
- Real-Time TDDFT<sup>6</sup>
- ROKS/CIS<sup>7</sup>

## Other Methods

- Static Exchange Approx. (IVO-HF)<sup>8</sup>
- Maximum Overlap ΔSCF<sup>9</sup>

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Accurate but Expensive

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More Economical but have limitations.

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Computationally Cheap,  
but inaccurate, requires  
energy shifts or SIC

## Functionals

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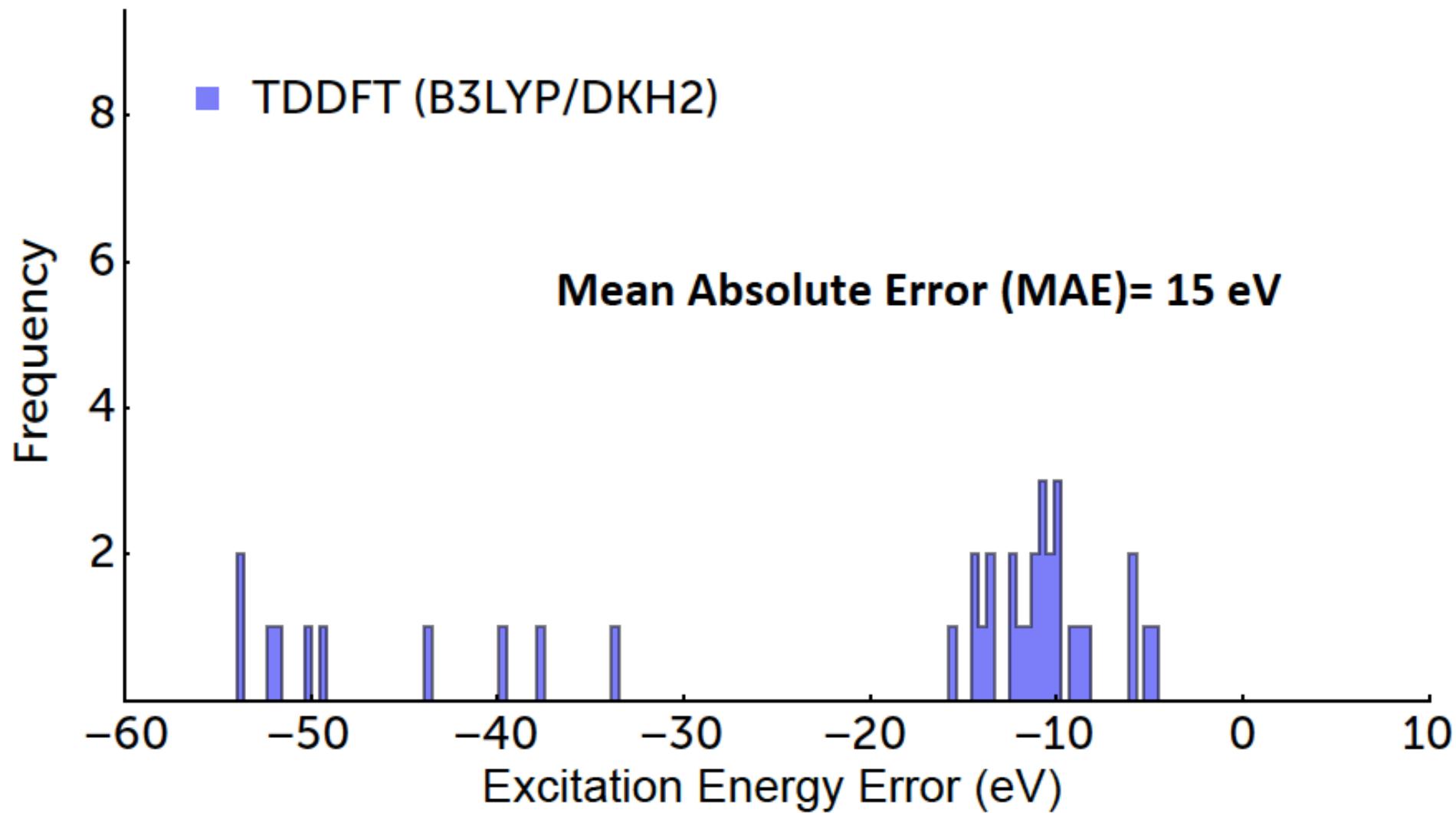
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# Core-Excited States With TDDFT



**TEST SET** → First-row elements: CO, H<sub>2</sub>CO, N<sub>2</sub>O, N<sub>2</sub>, HCN, CH<sub>4</sub>, C<sub>2</sub>H<sub>4</sub>  
Second-row elements: SiH<sub>4</sub>, PH<sub>3</sub>, H<sub>2</sub>S, SO<sub>2</sub>, HCl, Cl<sub>2</sub>

# Compensating for Failure of TDDFT

- Standard functionals fail dramatically for core excitations, which results in a large underestimation of the excitation energy.
  - Poor performance stems the low quality of the KS eigenvalues, use of frequency independent XC functionals.
- It is common for TDDFT spectra to be shifted by a uniform amount in order to compensate for the underestimation.

Method	Avg. Energy Shift (eV)
BP86	171.1
B3LYP	143.3
BP86	171.0
BP86/DKH2	105.6

# Orthogonality Constrained Density Functional Theory (OCDFT)

- Variational time-independent formulation of DFT
  - Builds upon variational DFT approaches<sup>1-7</sup> but imposes an orthogonality condition on the Kohn-Sham system:

$$\langle \Phi^{(m)} | \Phi^{(n)} \rangle = \delta_{mn}$$

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- OCDFT also introduces a quasi-adiabatic approximation in which the XC functional for each excited state is approximated as the ground state XC functional.

$$E_{\text{OCDFT}}^{(n)}[\rho^{(n)}] = \sum_{\mu\nu} D_{\mu\nu}^n (T_{\mu\nu} + V_{\mu\nu}) + E_{\text{coul}}[\rho^{(n)}] + E_{\text{xc}}^{(0)}[\rho^{(n)}].$$

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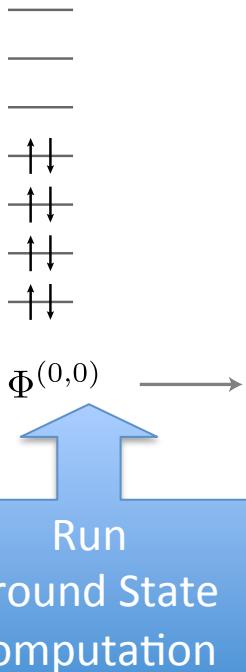
Adiabatic Approx.

# Extension of OCDFT to Simulate NEXAS Spectra

- Must introduce two new features to OCDFT to calculate NEXAS spectra
  - Previous implementation selected the *highest lying* hole orbitals (highest hole eigenvalue) however for core excitations we want to select the *lowest lying* hole orbitals (lowest hole eigenvalue).
  - Algorithm must be generalized to multiple excited states.

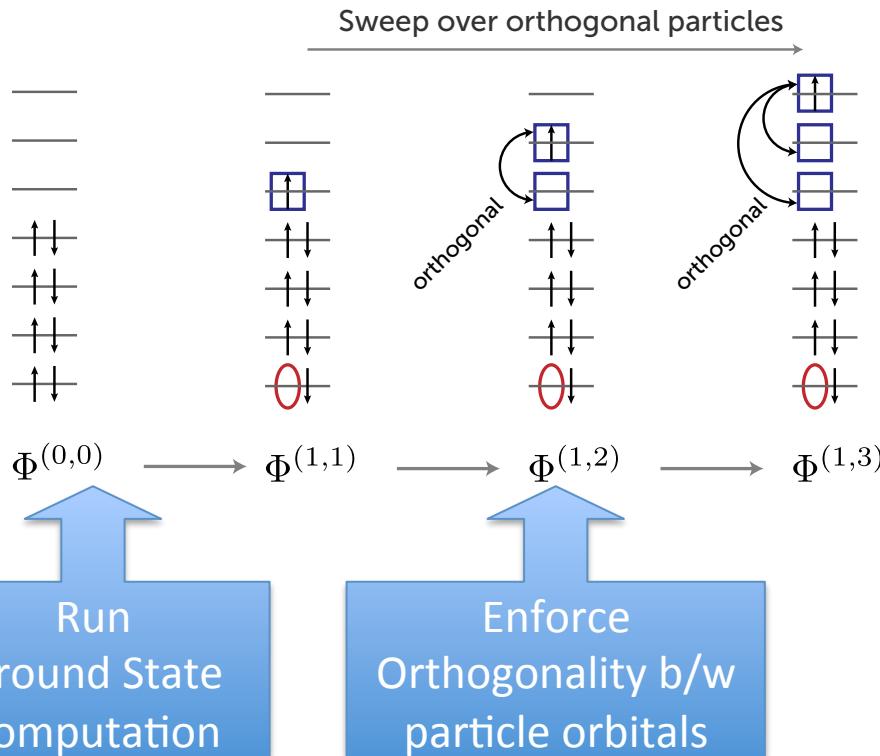
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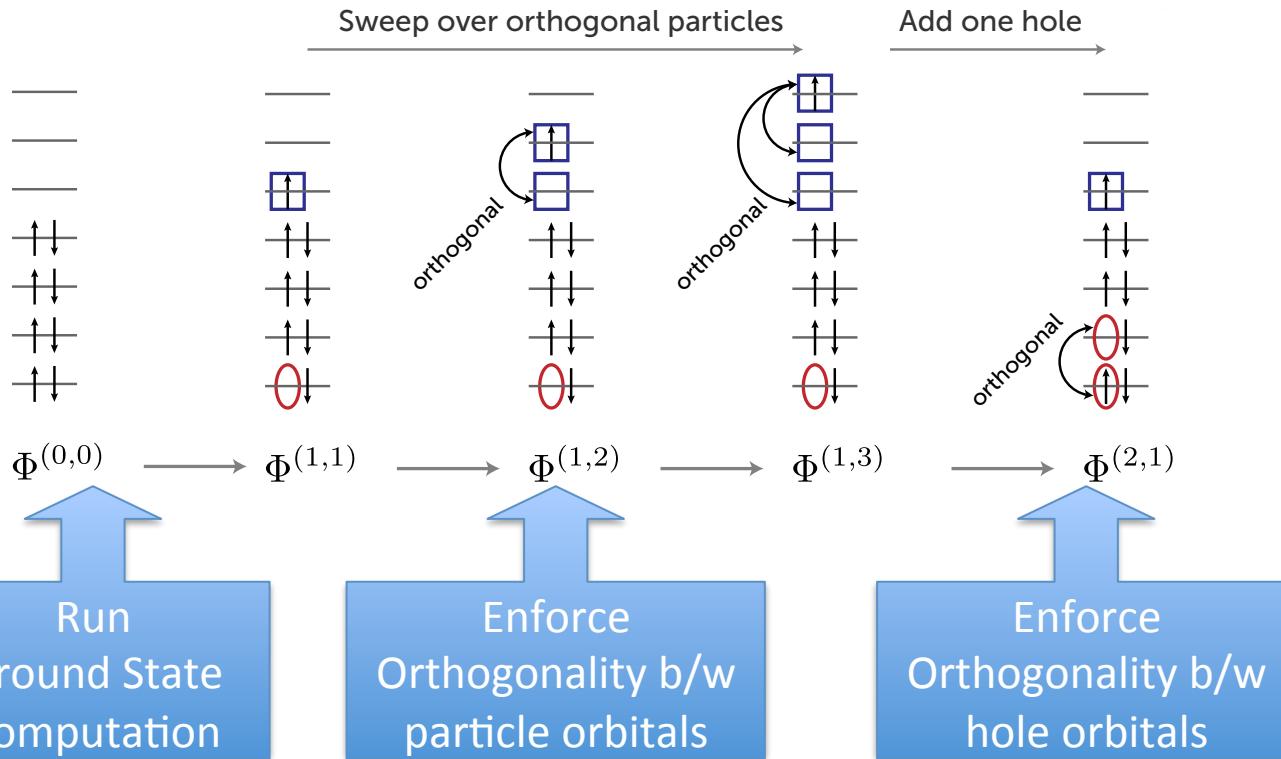
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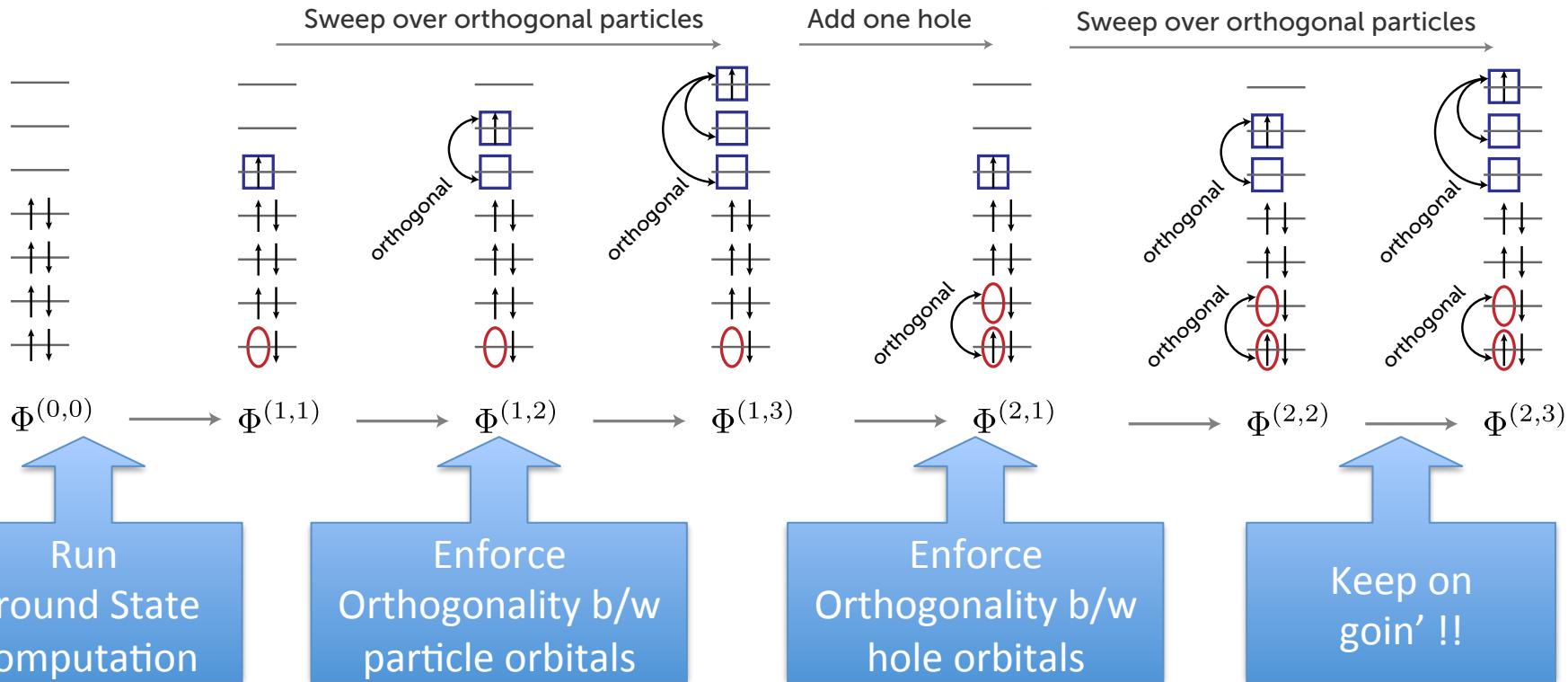
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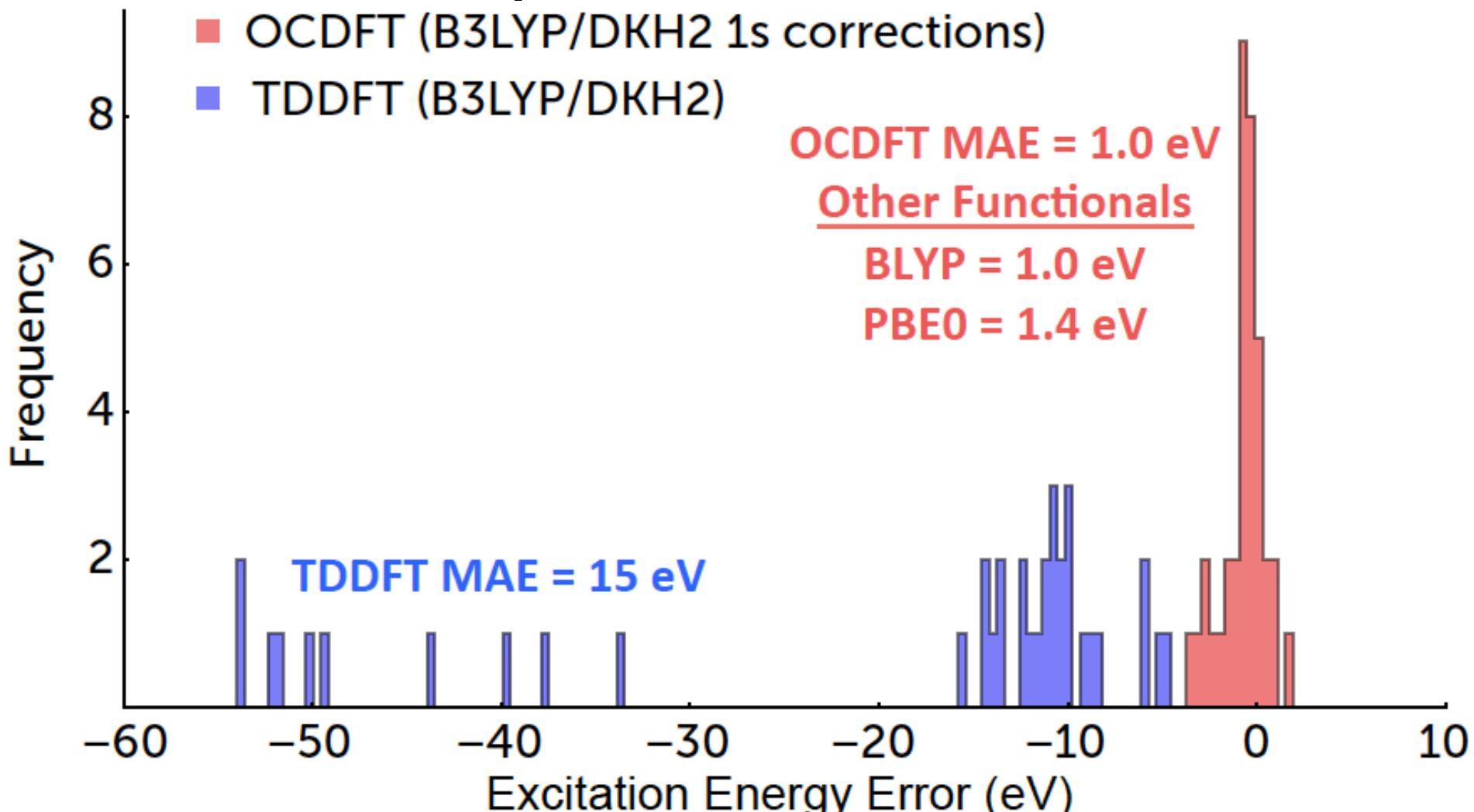


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# Comparison to TDDFT

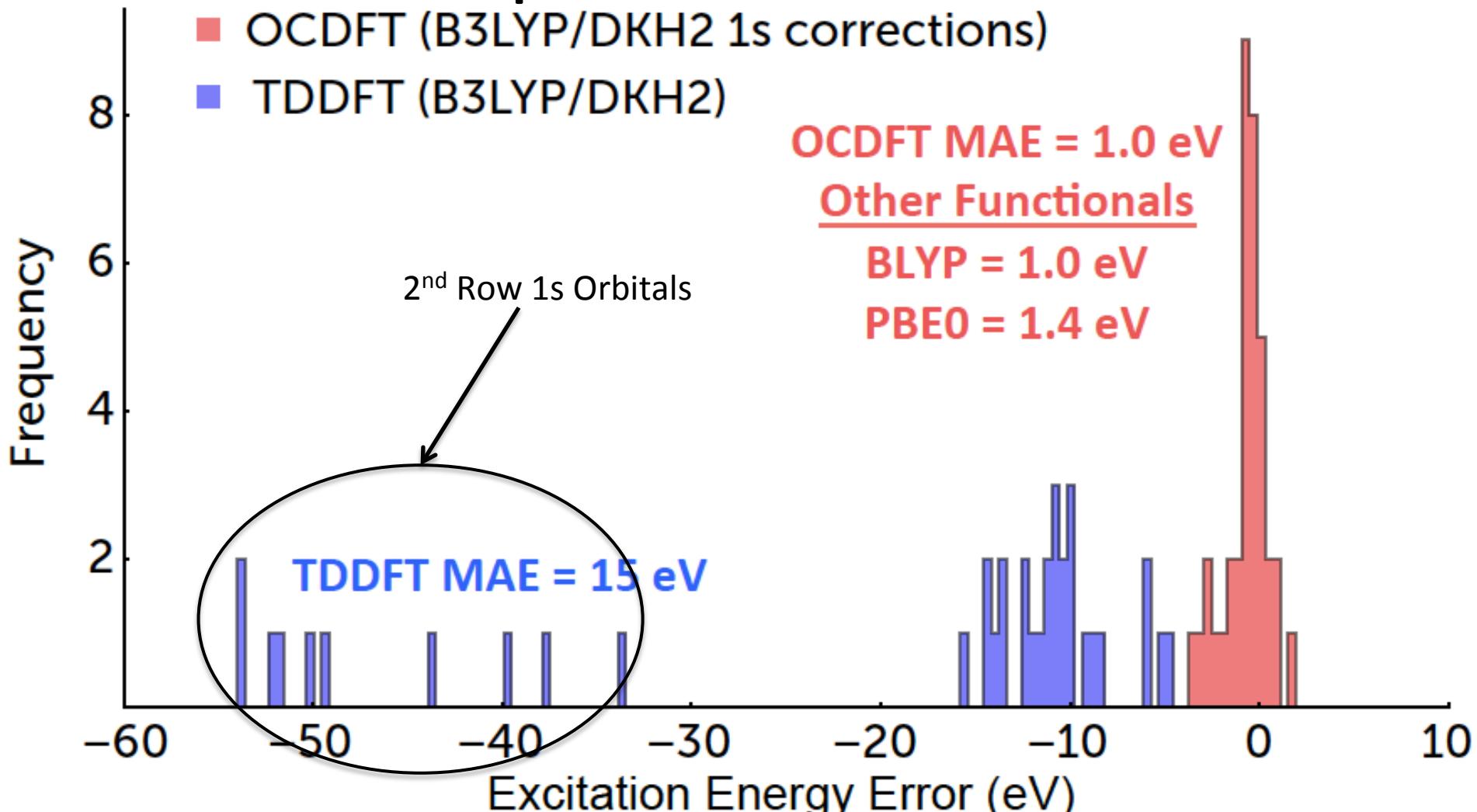


- EOM-CCSD<sup>1</sup> (MAE) = 0.9 eV
- SOS-CIS(D)<sup>2</sup> [MAE] = 1.2 eV
- Max Error for OCDFT: -3.7 eV
- Max Error for TDDFT: -53.6 eV

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# Sensitivity to Orbital Overlap

- Comparing the sensitivity of OCDFT and TDDFT using Tozer Overlap Metric.<sup>1</sup>

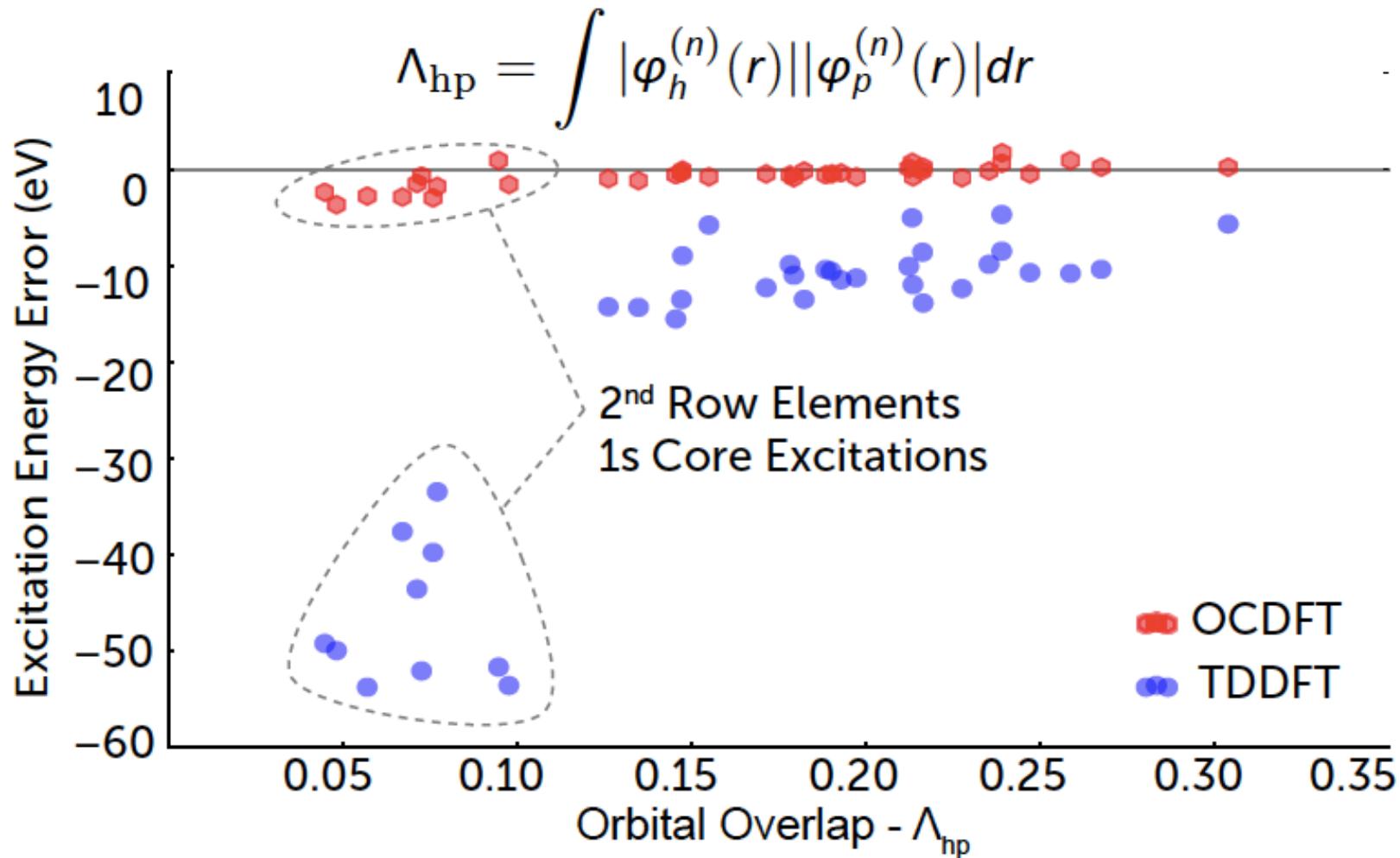
$$\Lambda_{hp} = \int |\varphi_h^{(n)}(r)| |\varphi_p^{(n)}(r)| dr$$

-

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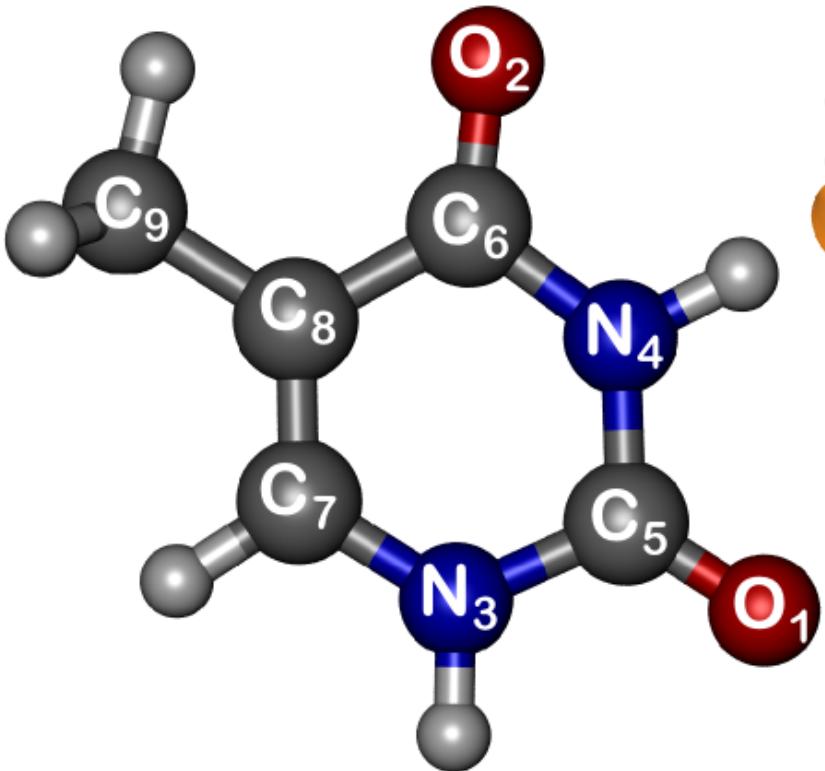
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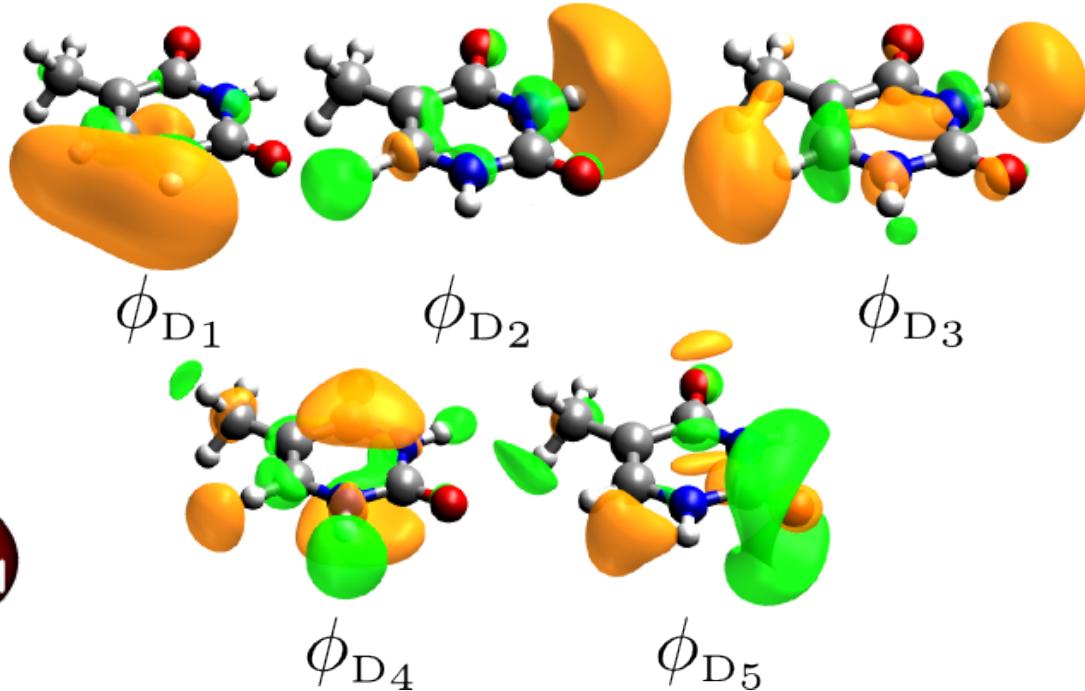


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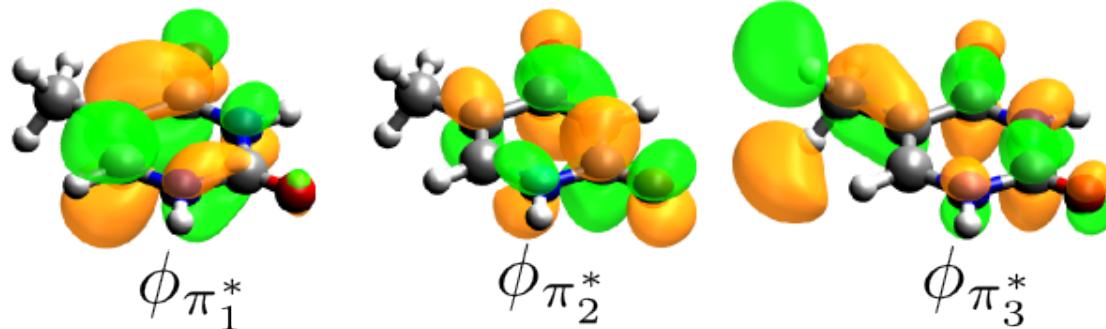
# Application to Thymine



Diffuse Orbitals



Anti-Bonding Orbitals



# Spectral Simulation Details

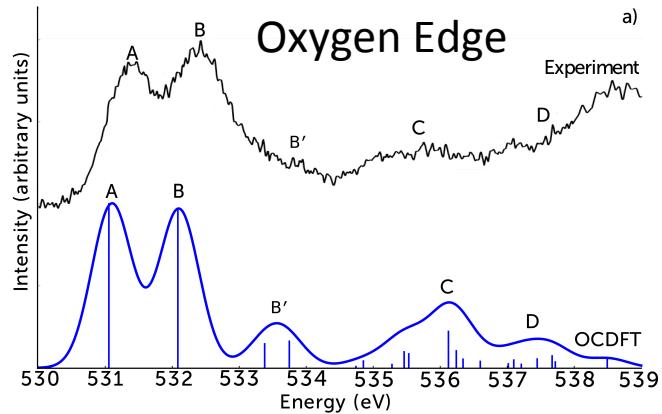
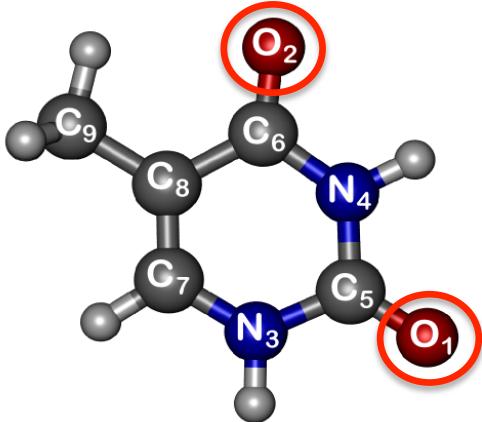
- Compute 10 excitations per hole for each carbon, nitrogen, and oxygen 1s orbital in thymine.
- Plot spectra using gaussians with FWHM of 0.1 eV - 0.4 eV in order to simulate natural spectroscopic broadening effects.
- Compare the OCDFT results with experiment and previously applied 2nd order algebraic diagrammatic construction [ADC(2)] methods
- Transition dipole moments are approximated using Kohn-Sham determinants and the position vector

$$\mu_{fi} = \langle \Phi^{(f)} | \hat{r} | \Phi^{(i)} \rangle$$

- Using the approximate transition dipole moment we can now calculate an oscillator strength for each transition

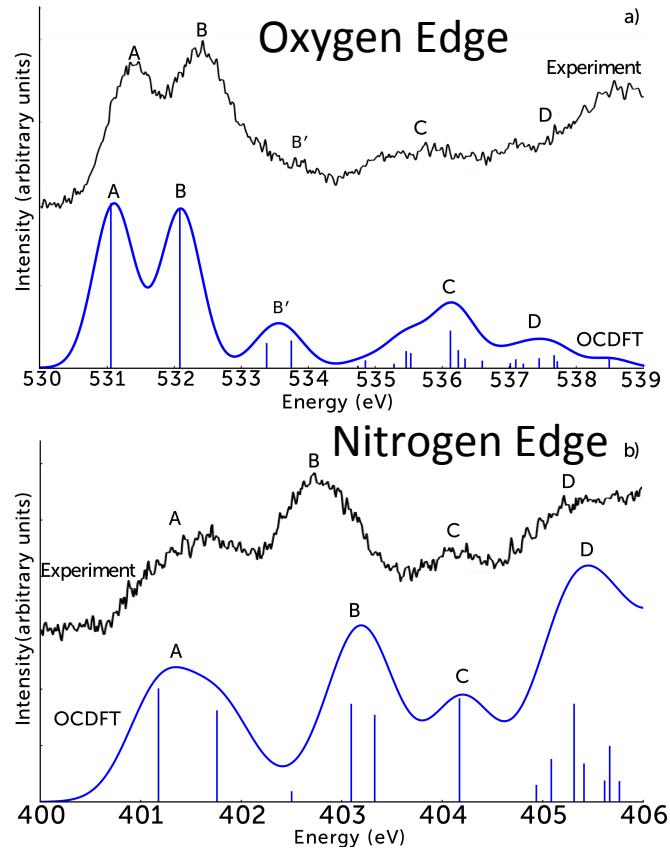
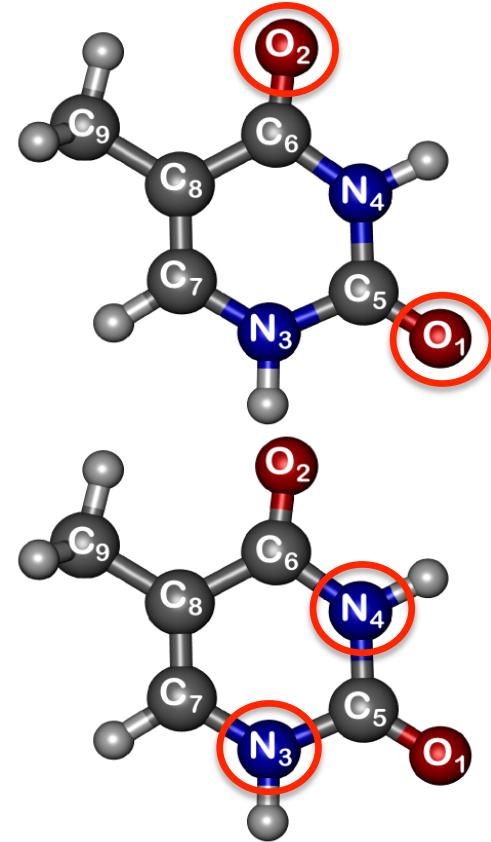
$$f_{\text{osc}} = \frac{2}{3} |\mu_{fi}|^2 \omega_{fi}$$

# OCDFT Simulation of Thymine K-Edge



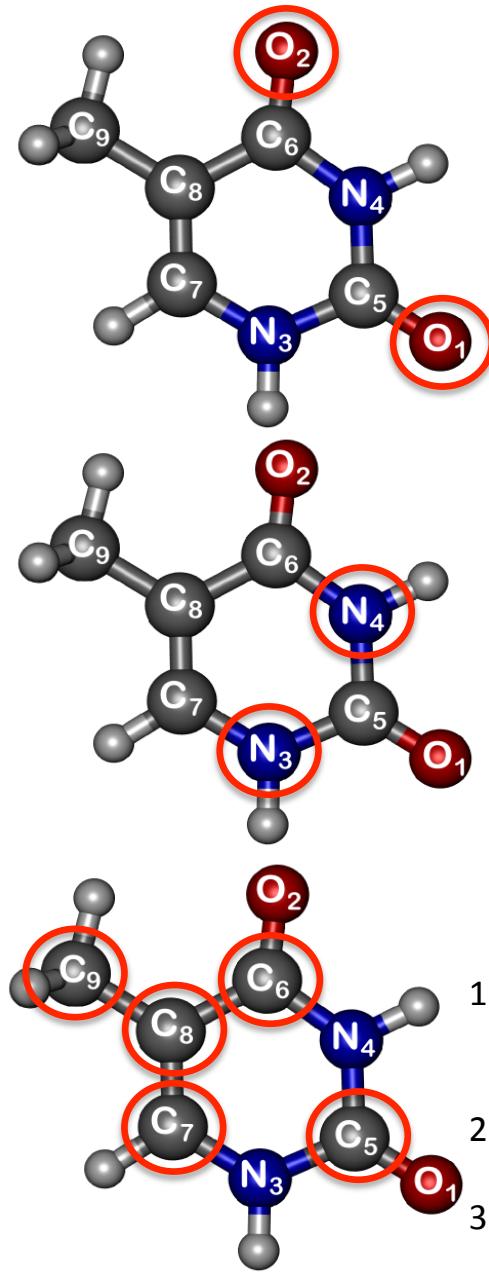
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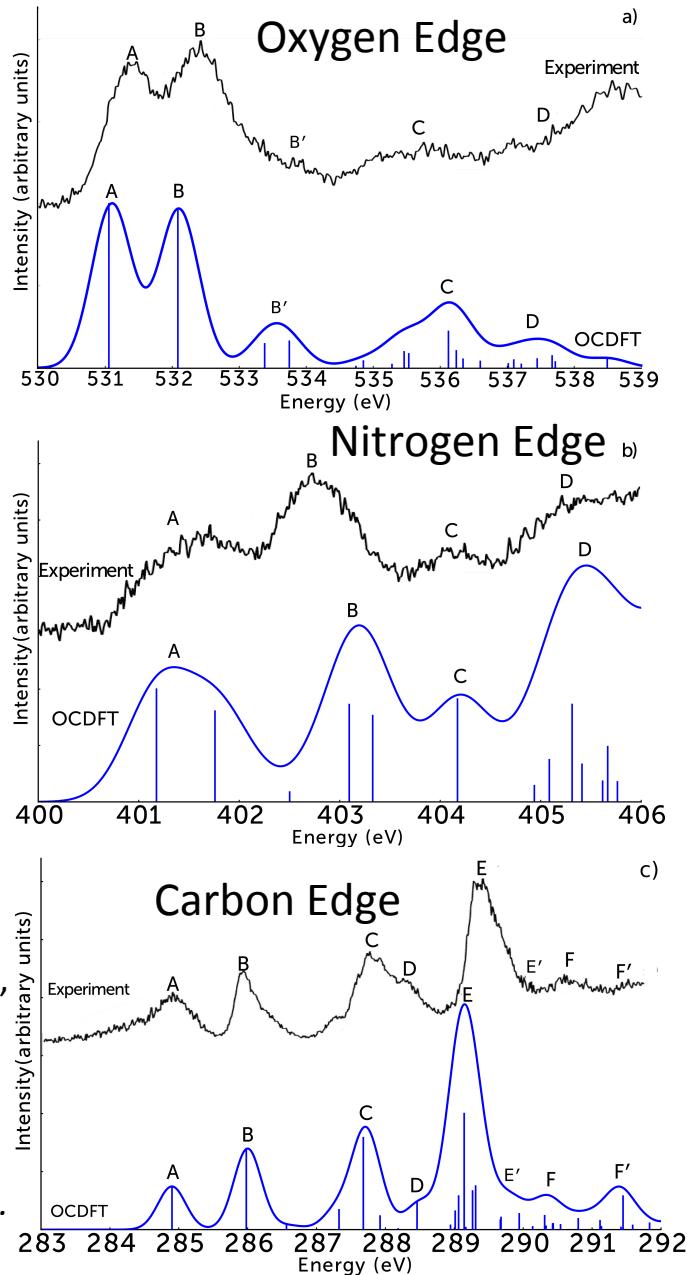


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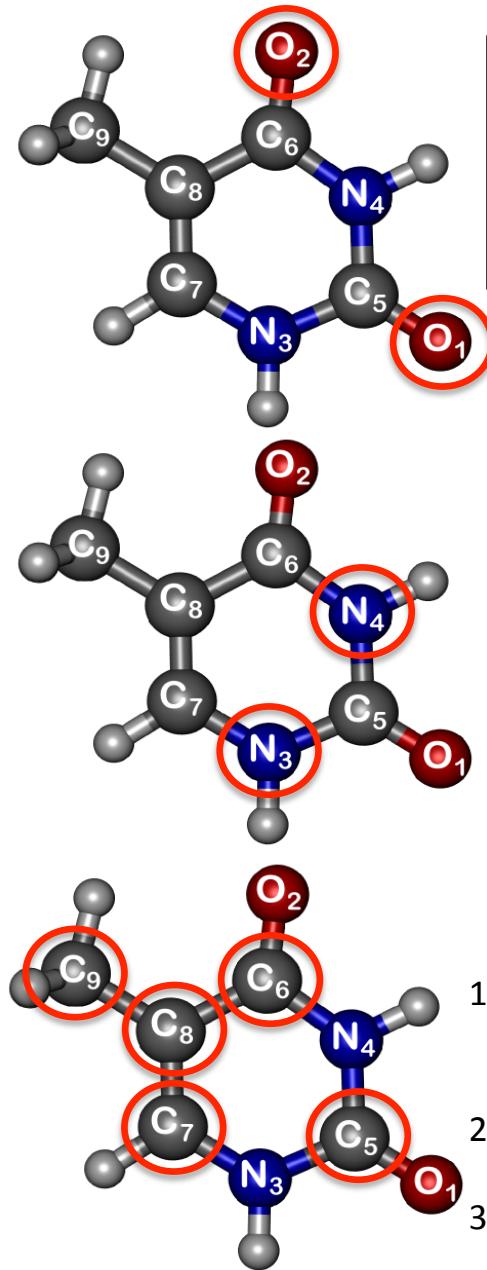
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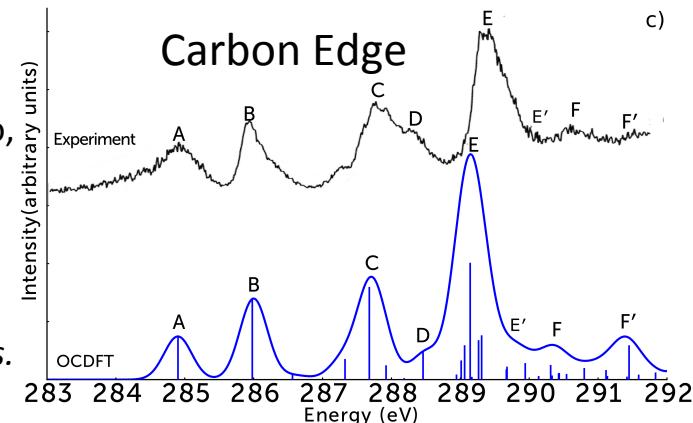
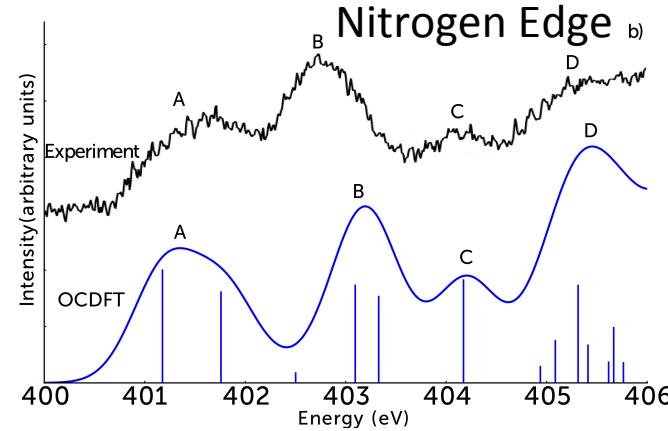
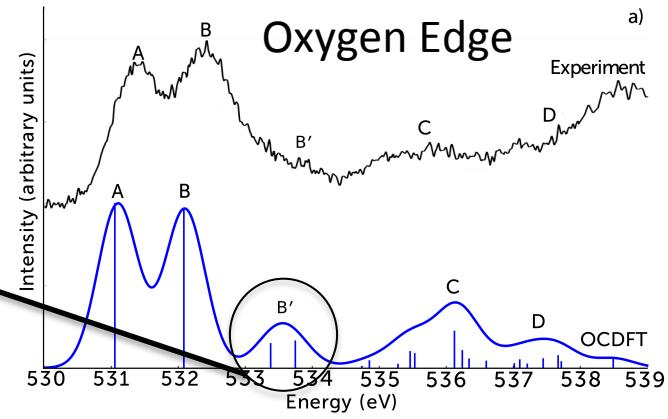
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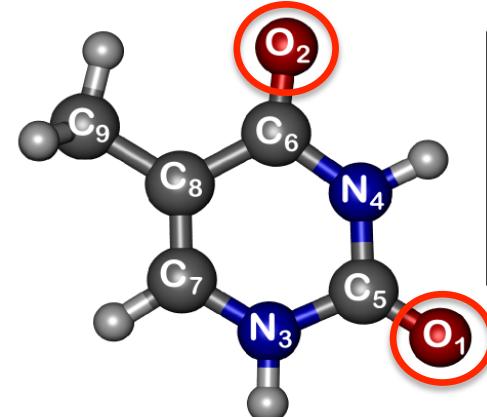


Shoulder Feature B' initially absent in ADC(2) spectrum.<sup>1</sup>  
Was confirmed by later CVS-ADC(2) study.<sup>2</sup>

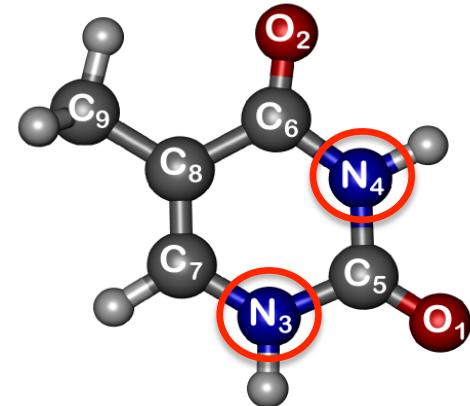


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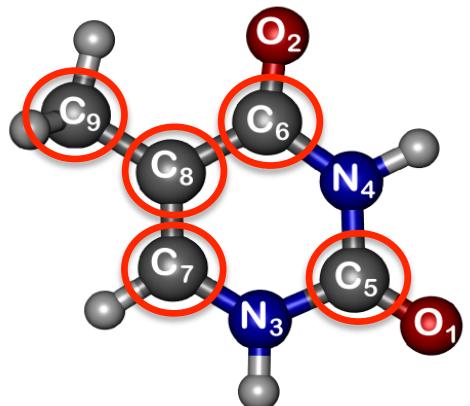
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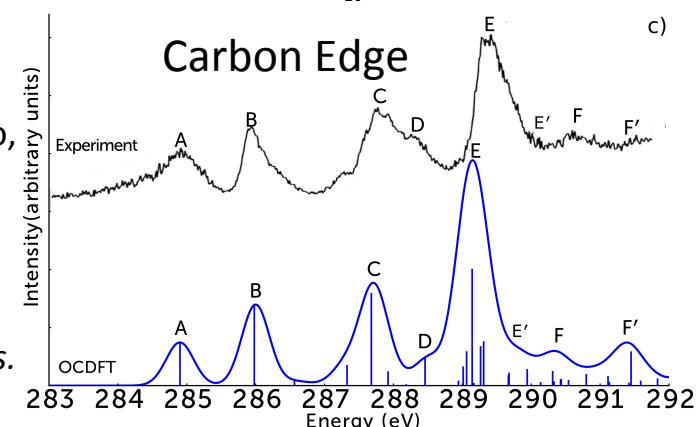
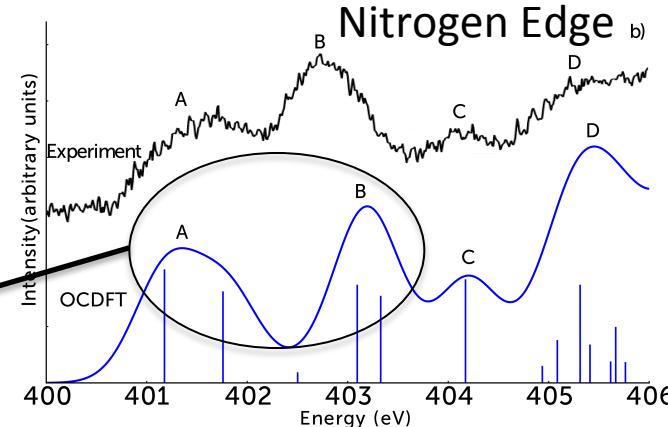
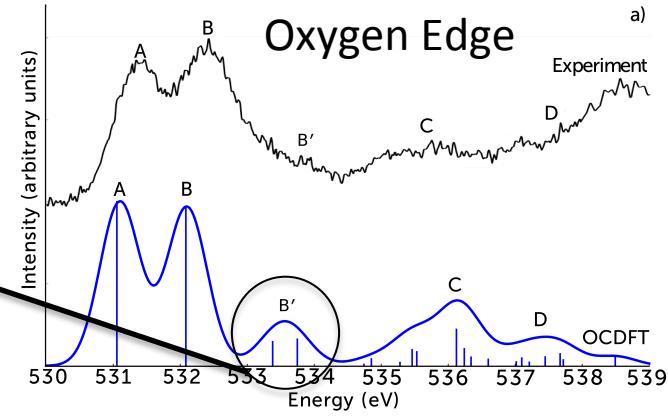
Shoulder Feature B' initially absent in ADC(2) spectrum.<sup>1</sup>  
Was confirmed by later CVS-ADC(2) study.<sup>2</sup>



Peak Features A and B are unresolved in ADC(2) spectrum. Appear as single peak feature.

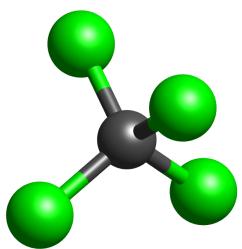


- 1) Plekan, O.; Feyer, V.; Richter, R.; Coreno, M.; de Simone, M.; et al. *Chem. Phys.* 2008
- 2) Wenzel, J.; Wormit, M.; Dreuw, A. J. *Comput. Chem.* 2014
- 3) Derricotte, W. D.; Evangelista, F. A. *Phys. Chem. Chem. Phys.* 2015

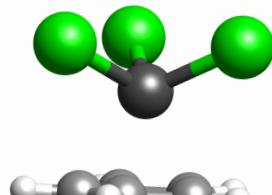


# Application to Transition Metal Complexes

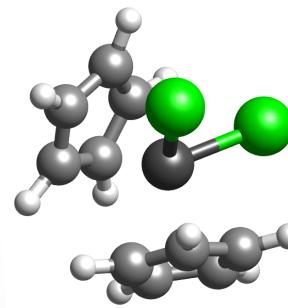
# Ti K-Edge of Tetra coordinated Titanium Complexes



$\text{TiCl}_4$



$\text{TiCpCl}_3$

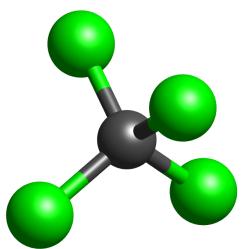


$\text{TiCp}_2\text{Cl}_2$

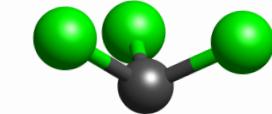
- Class of molecules used to study covalency in cyclopentadienyl (Cp) complexes relevant for use as anti-cancer drug.
- Full treatment of scalar relativistic effects will be handled by X2C Hamiltonian implemented as plugin in PSI4

- 1) DeBeer George, S.; Brant, P.; Solomon, E. I. *J. Am. Chem. Soc.* **2005**
- 2) Casarin, M.; Finetti, P.; Vittadini, A.; Wang, F.; Ziegler, T. *J. Phys. Chem. A* **2007**.

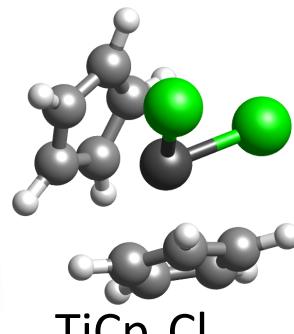
# Ti K-Edge of Tetra coordinated Titanium Complexes



$\text{TiCl}_4$

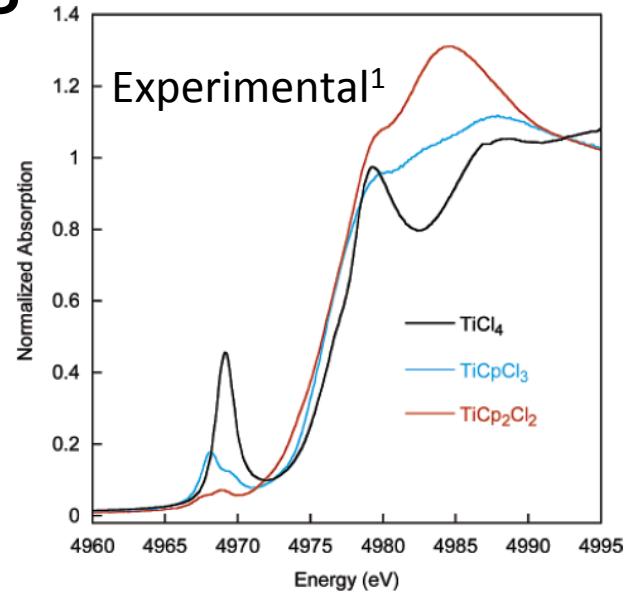


$\text{TiCpCl}_3$



$\text{TiCp}_2\text{Cl}_2$

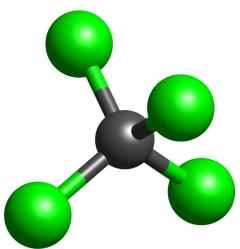
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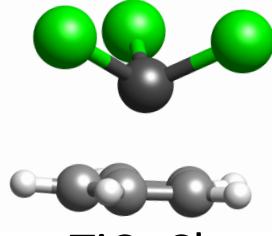
1) DeBeer George, S.; Brant, P.; Solomon, E. I. *J. Am. Chem. Soc.* **2005**

2) Casarin, M.; Finetti, P.; Vittadini, A.; Wang, F.; Ziegler, T. *J. Phys. Chem. A* **2007**.

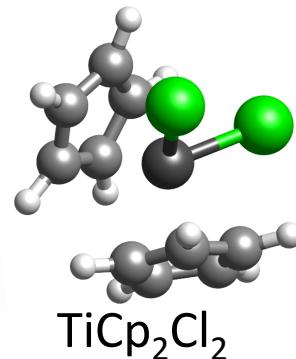
# Ti K-Edge of Tetra coordinated Titanium Complexes



$\text{TiCl}_4$

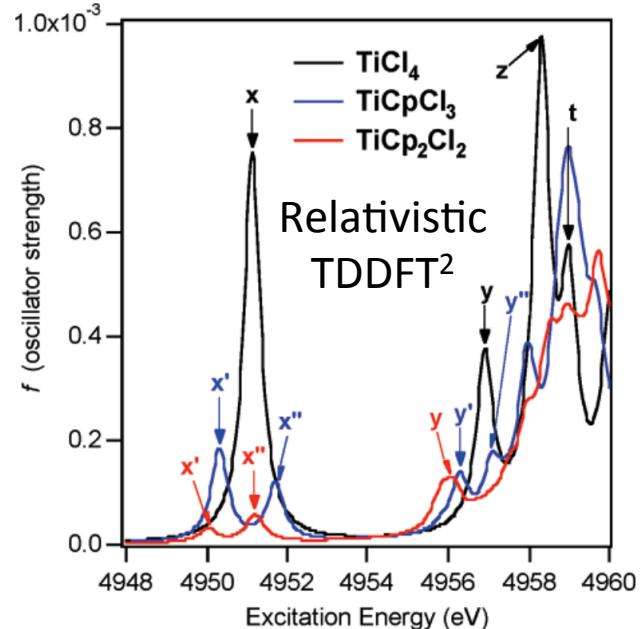
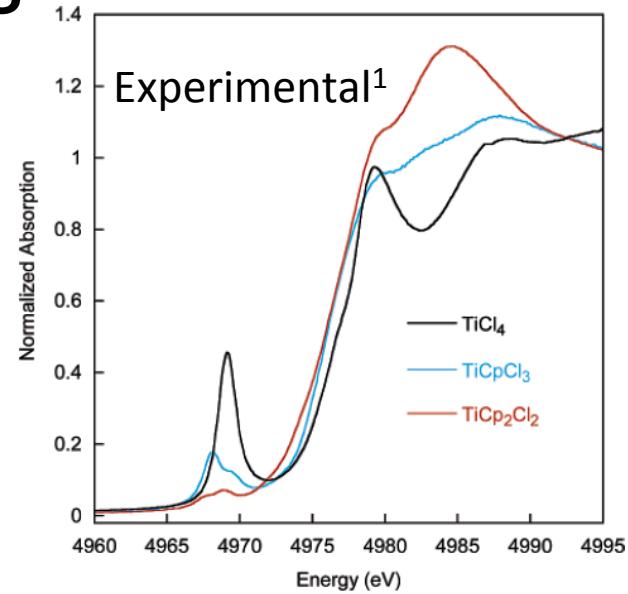


$\text{TiCpCl}_3$



$\text{TiCp}_2\text{Cl}_2$

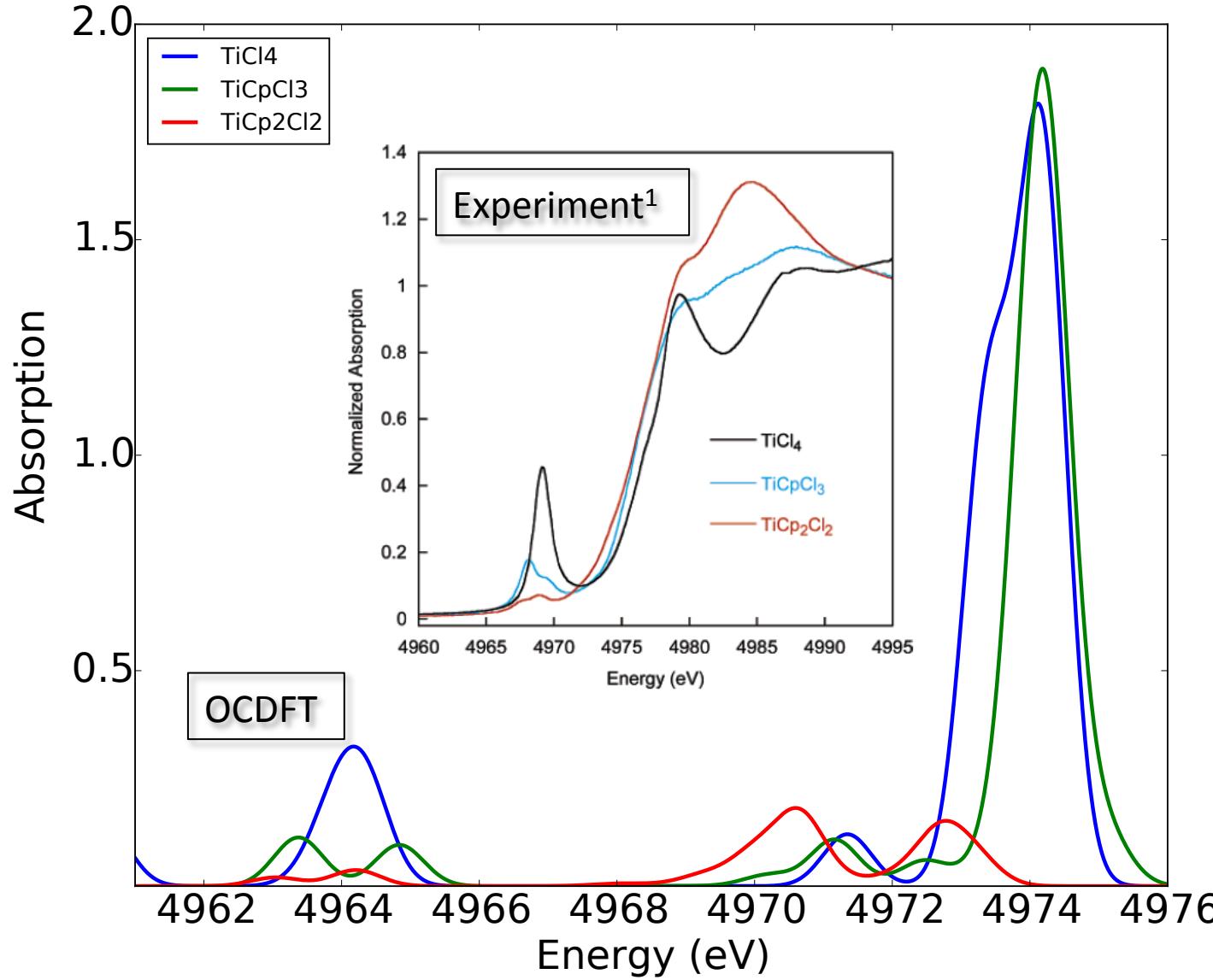
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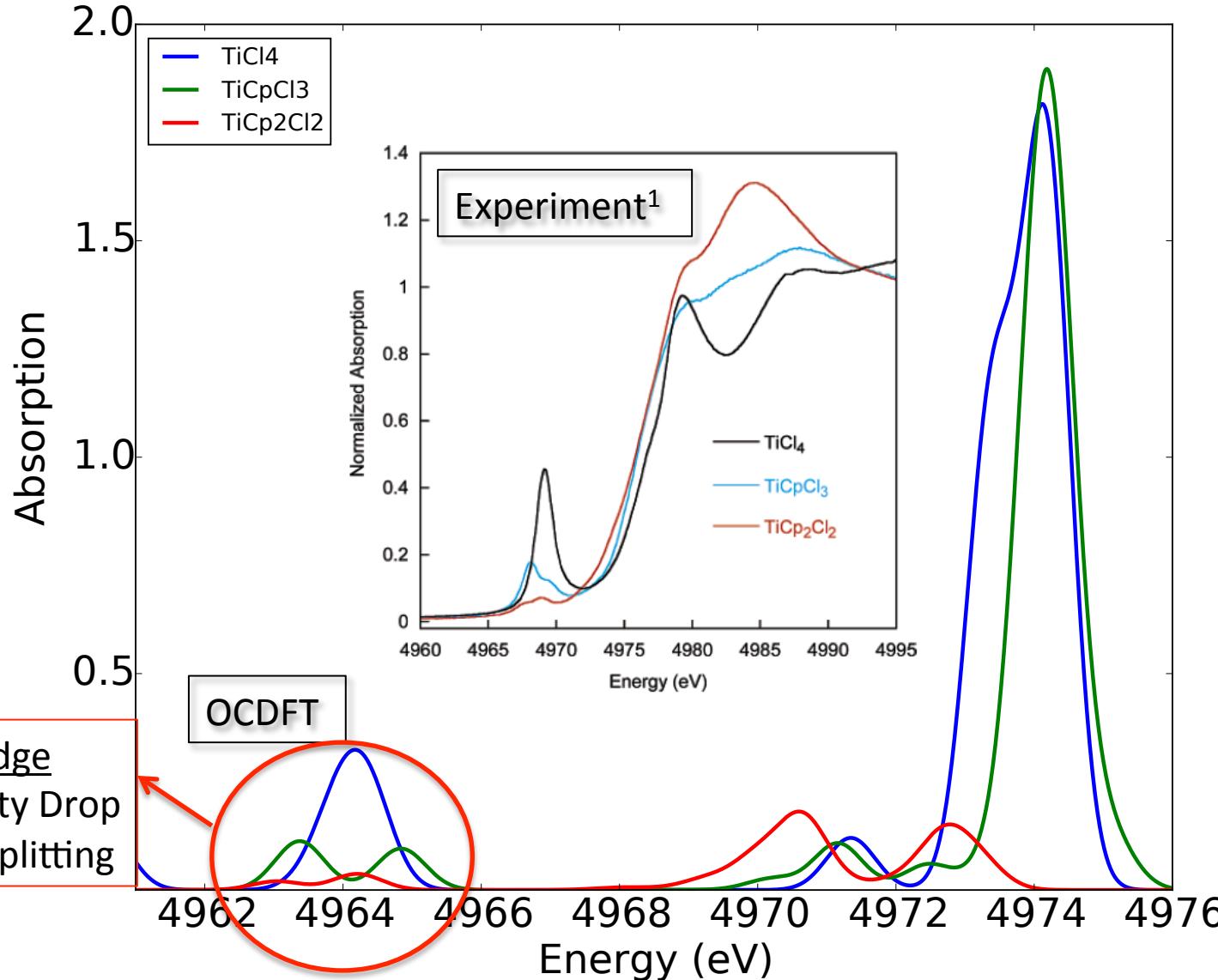
# Comparison With Experimental Spectra



- Why are the peaks split for the Cp compounds? Maybe OCDFT Particle Orbitals can shed some insight?

1) DeBeer George, S.; Brant, P.; Solomon, E. I. *J. Am. Chem. Soc.* 2005

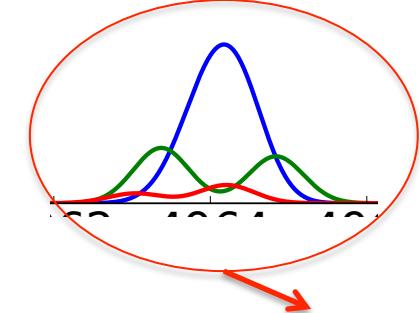
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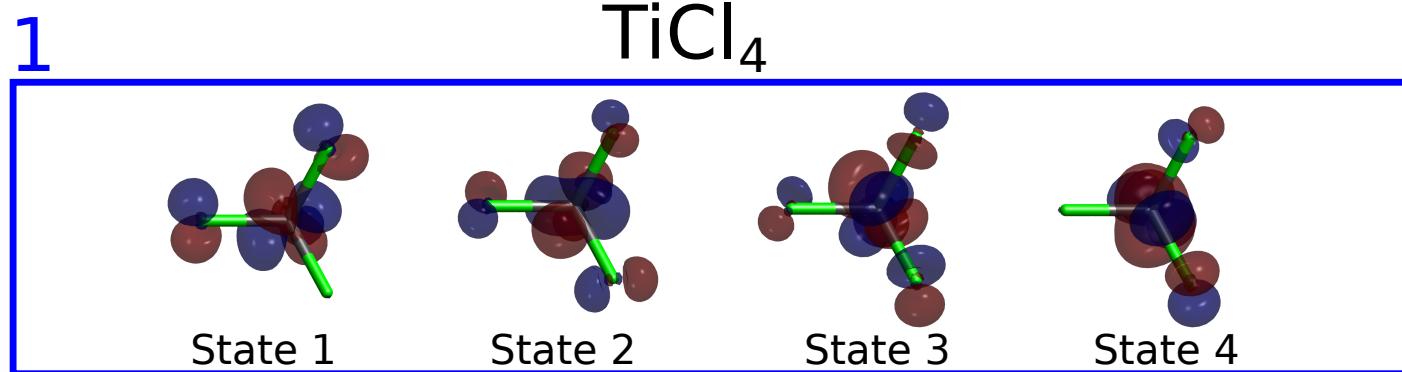
1) DeBeer George, S.; Brant, P.; Solomon, E. I. *J. Am. Chem. Soc.* 2005

# Pre-Edge Analysis

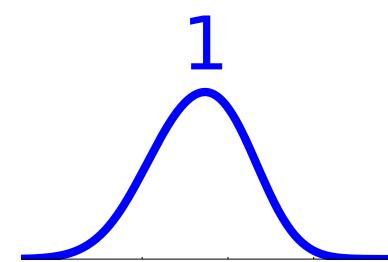


State	Transition Energy (eV)	Rel. Osc. Strength
$\text{TiCl}_4$		
1	4963.85	0.0805
2	4963.53	0.0181
3	4964.38	0.1983
4	4964.03	0.1120

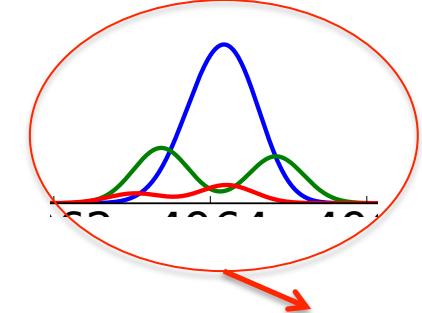
## *Particle Orbitals*



## *Pre-Edge*

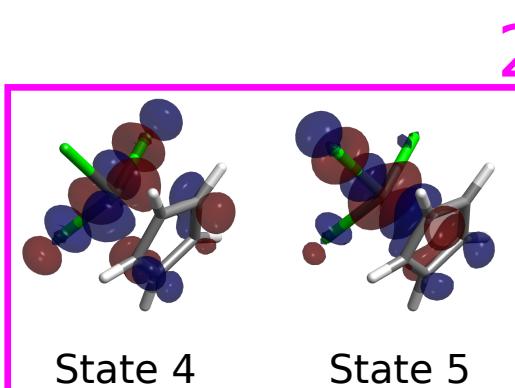
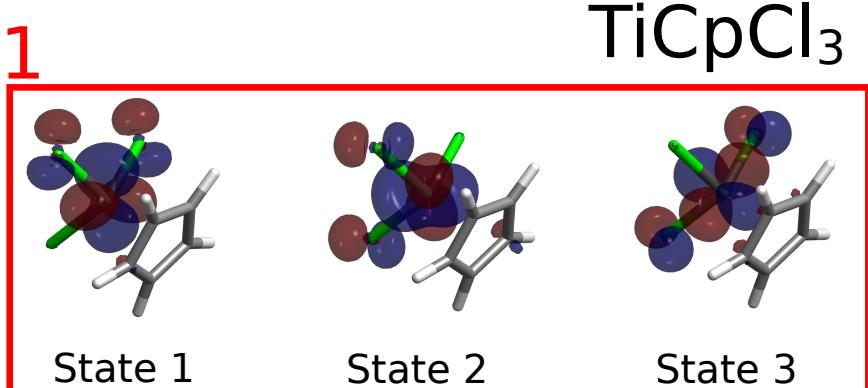


# Pre-Edge Analysis



State	Transition Energy (eV)	Rel. Osc. Strength
<b>TiCpCl<sub>3</sub></b>		
1	4963.31	0.0226
2	4963.38	0.0227
3	4963.45	0.0190
4	4964.80	0.0271
5	4964.88	0.0270

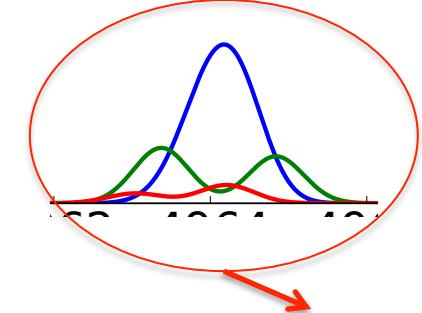
## *Particle Orbitals*



## *Pre-Edge*

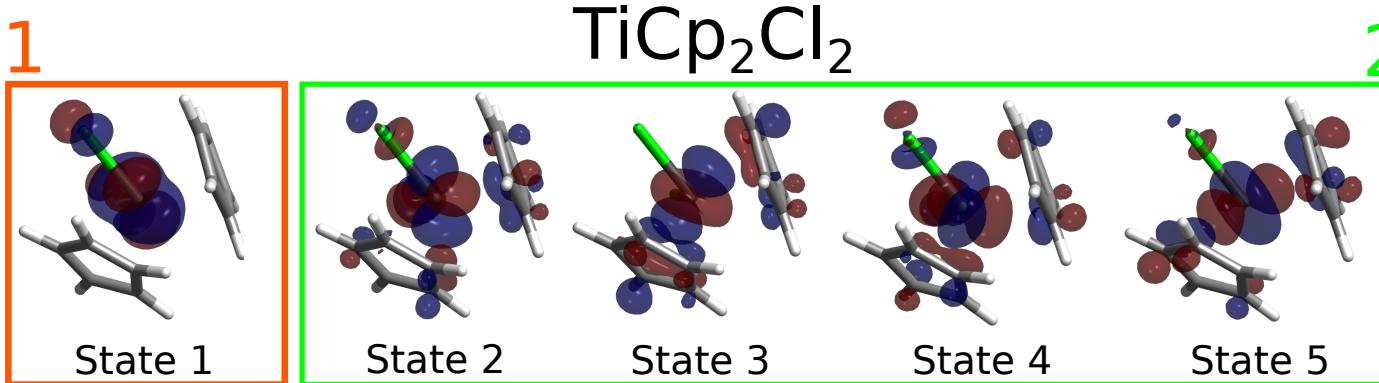


# Pre-Edge Analysis

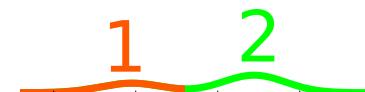


State	Transition Energy (eV)	Rel. Osc. Strength
<b>TiCp<sub>2</sub>Cl<sub>2</sub></b>		
1	4963.04	0.0114
2	4964.20	0.0079
3	4964.18	0.0012
4	4964.32	0.0003
5	4964.22	0.0116

## Particle Orbitals



## Pre-Edge



# Conclusions

- Extended OCDFT to calculate multiple excited states in order to fully simulate NEXAS spectra.
- OCDFT has no significant dependence on the amount of Hartree-Fock exchange present in the functional
- Looked at excitations from first and second row elements and show that OCDFT is less sensitive to changes in orbital overlap.
- Calculated the NEXAS spectra of thymine in order to show that OCDFT is a useful tool for interpreting NEXAS spectra.
- Coupled OCDFT with X2C relativistic Hamiltonian in order to show that OCDFT can be effective at simulating NEXAS spectra of transition metal complexes.

# Thank You



## Evangelista Lab Members

Francesco Evangelista ← Advisor

Prakash Verma  
Chenyang Li

Kevin Hannon  
Jeff Schriber  
Tianyuan Zhang

← Post-Docs

← Minions



National Institutes  
of Health

