



EMORY
UNIVERSITY



Recent Applications of Orthogonality Constrained Density Functional Theory: Core Excitations

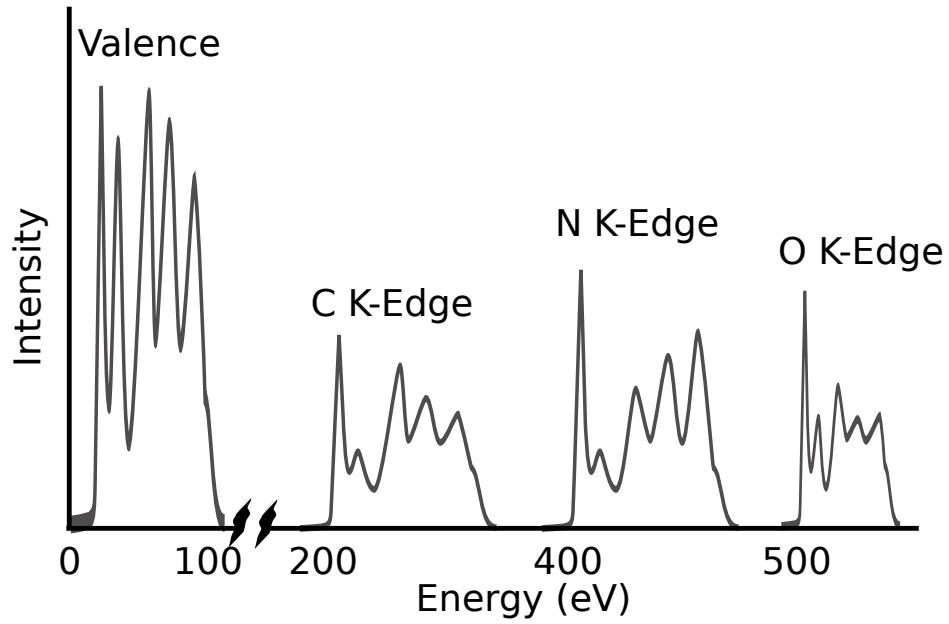
Wallace Derricotte, Prakash Verma, Francesco Evangelista

NOBCChE Presentation

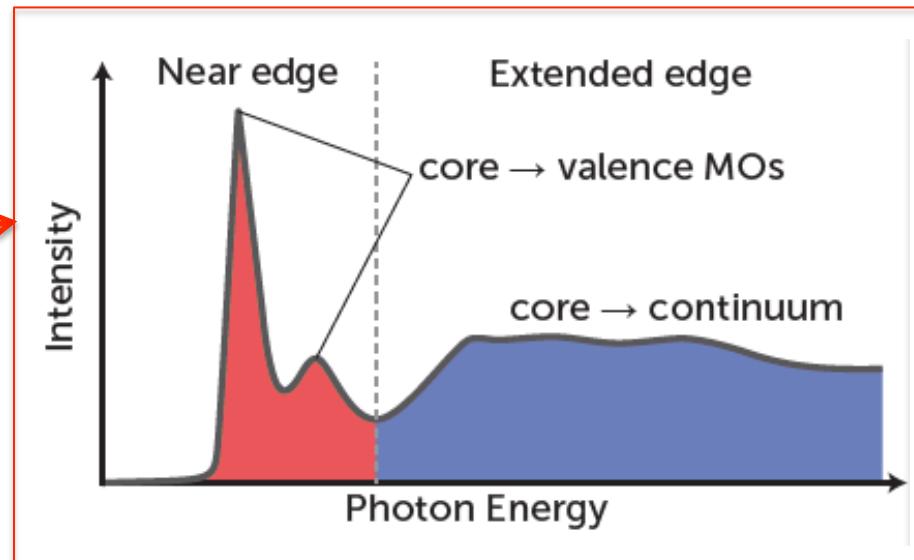
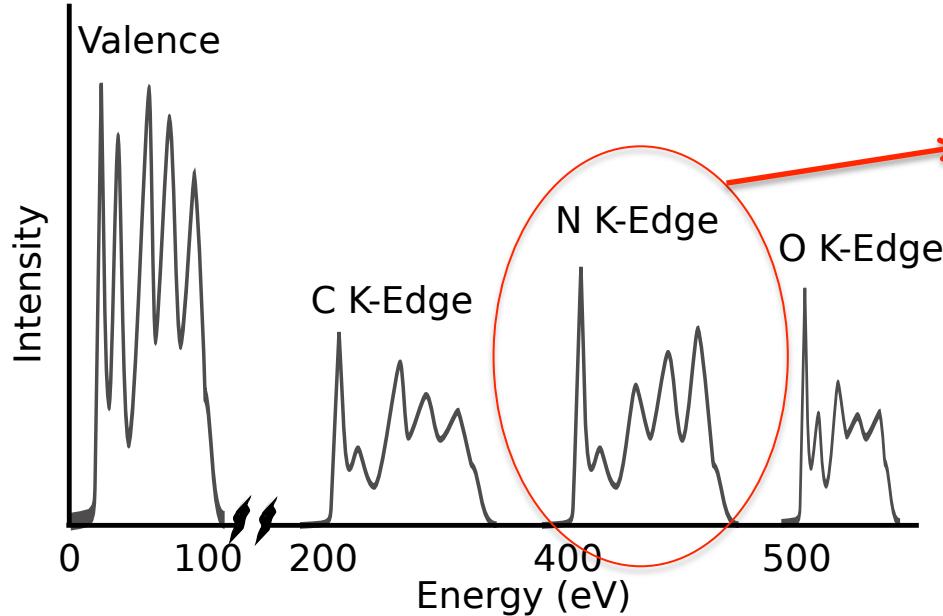
Evangelista Lab

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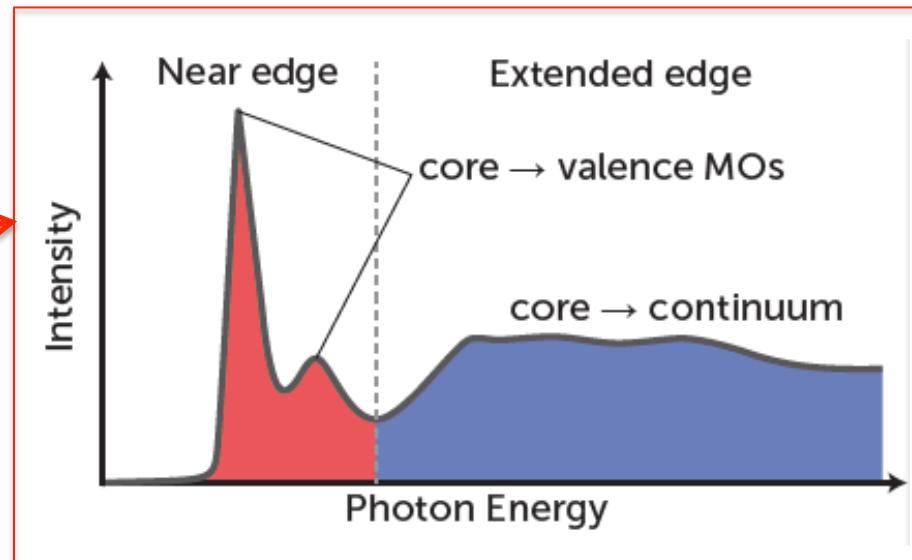
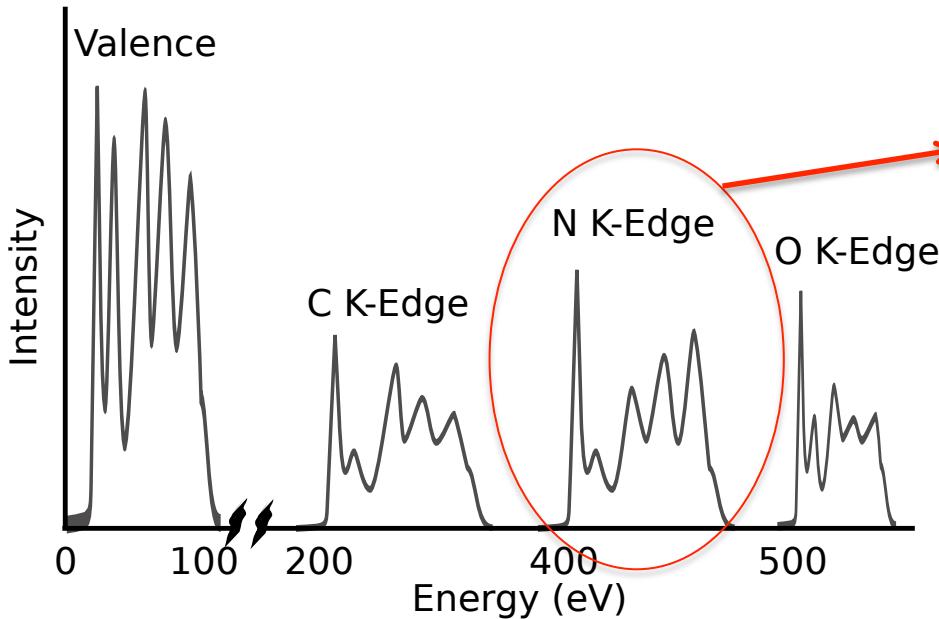
Core-Excited States



Core-Excited States



Core-Excited States



- Theoretical Challenges Include:
 - Orbital relaxation
 - Require treatment of scalar (1s excitations) and spin-dependent relativistic effects
 - Electron correlation

Computational Approaches

Wavefunction/Green's Function Methods

- SOS-CIS(D)¹
- RASSCF²
- LR-CC methods³
- Algebraic Diagrammatic Construction (ADC)⁴

TDDFT

- Linear Response TDDFT⁵
- Real-Time TDDFT⁶
- ROKS/CIS⁷

Other Methods

- Static Exchange Approx. (IVO-HF)⁸
- Maximum Overlap ΔSCF⁹

1) Asmurf, F.A.; Besley, N.A., *Chem. Phys. Lett.* **2008**

2) Ågren, H.; Jensen, H.J.A.; *Chem. Phys.* **1993**

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Computational Approaches

Wavefunction/Green's Function Methods

- SOS-CIS(D)¹
- RASSCF²
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- Algebraic Diagrammatic Construction (ADC)⁴

Accurate but Expensive

TDDFT

- Linear Response TDDFT⁵
- Real-Time TDDFT⁶
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Computational Approaches

Wavefunction/Green's Function Methods

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More Economical but have limitations.

TDDFT

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Computational Approaches

Wavefunction/Green's Function Methods

- SOS-CIS(D)¹
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- Algebraic Diagrammatic Construction (ADC)⁴

Computationally Cheap,
but inaccurate, requires
energy shifts or SIC

Functionals

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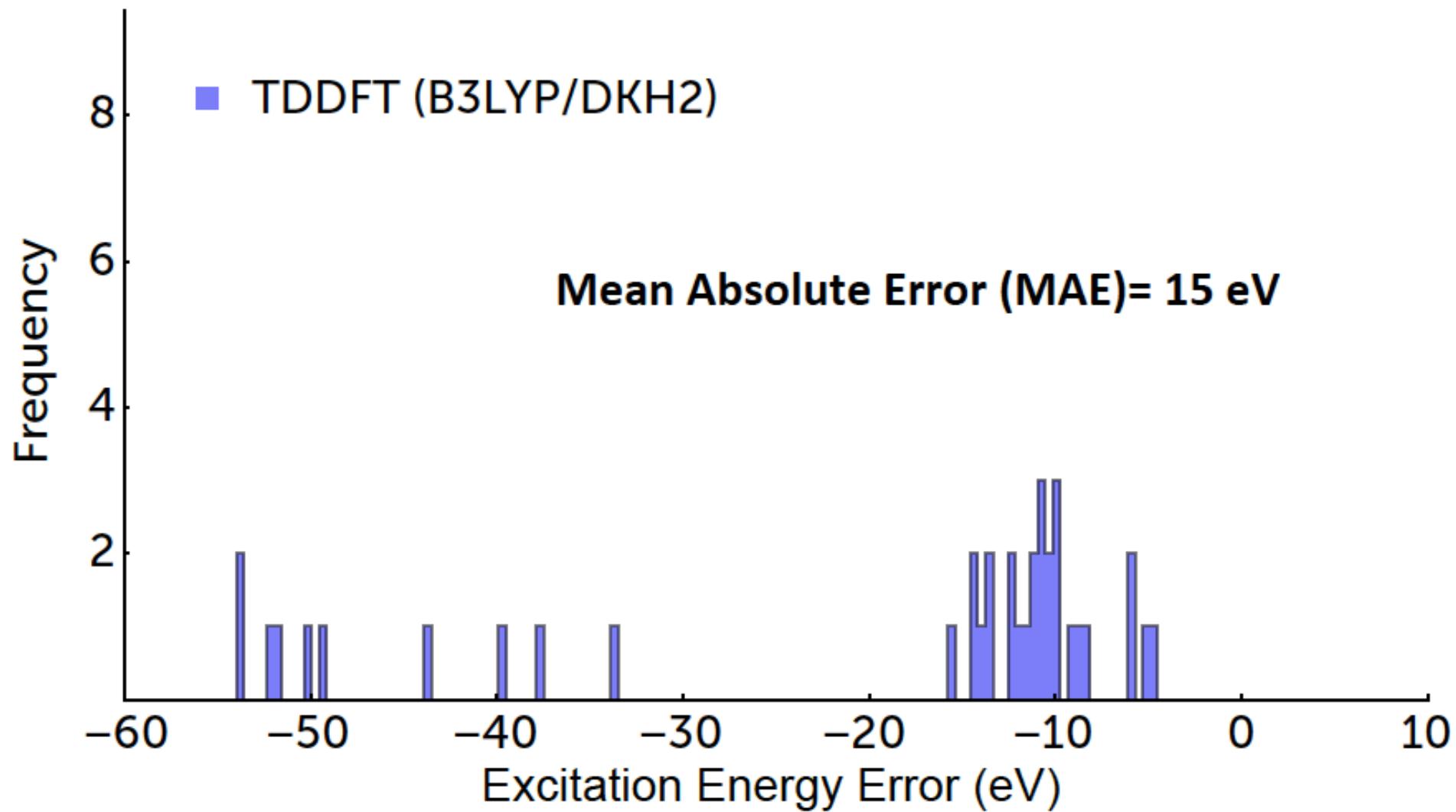
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Core-Excited States With TDDFT



TEST SET → First-row elements: CO, H₂CO, N₂O, N₂, HCN, CH₄, C₂H₄
Second-row elements: SiH₄, PH₃, H₂S, SO₂, HCl, Cl₂

Compensating for Failure of TDDFT

- Standard functionals fail dramatically for core excitations, which results in a large underestimation of the excitation energy.
 - Poor performance stems the low quality of the KS eigenvalues, use of frequency independent XC functionals.
- It is common for TDDFT spectra to be shifted by a uniform amount in order to compensate for the underestimation.

Method	Avg. Energy Shift (eV)
BP86	171.1
B3LYP	143.3
BP86	171.0
BP86/DKH2	105.6

Orthogonality Constrained Density Functional Theory (OCDFT)

- Variational time-independent formulation of DFT
 - Builds upon variational DFT approaches¹⁻⁷ but imposes an orthogonality condition on the Kohn-Sham system:

$$\langle \Phi^{(m)} | \Phi^{(n)} \rangle = \delta_{mn}$$

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- 2) Kowalczyk, T.; Yost, S. R.; Van Voorhis, T. J. *Chem. Phys.* **2011**
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- OCDFT also introduces a quasi-adiabatic approximation in which the XC functional for each excited state is approximated as the ground state XC functional.

$$E_{\text{OCDFT}}^{(n)}[\rho^{(n)}] = \sum_{\mu\nu} D_{\mu\nu}^n (T_{\mu\nu} + V_{\mu\nu}) + E_{\text{coul}}[\rho^{(n)}] + E_{\text{xc}}^{(0)}[\rho^{(n)}].$$

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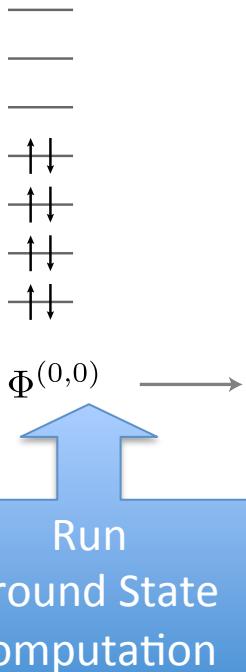
Adiabatic Approx.

Extension of OCDFT to Simulate NEXAS Spectra

- Must introduce two new features to OCDFT to calculate NEXAS spectra
 - Previous implementation selected the *highest lying* hole orbitals (highest hole eigenvalue) however for core excitations we want to select the *lowest lying* hole orbitals (lowest hole eigenvalue).
 - Algorithm must be generalized to multiple excited states.

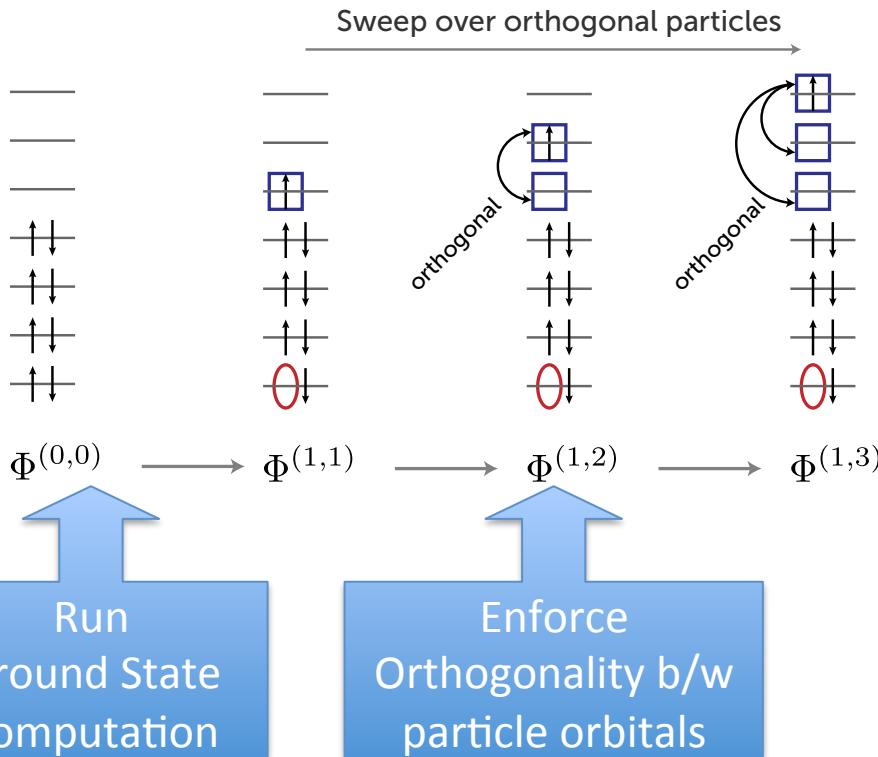
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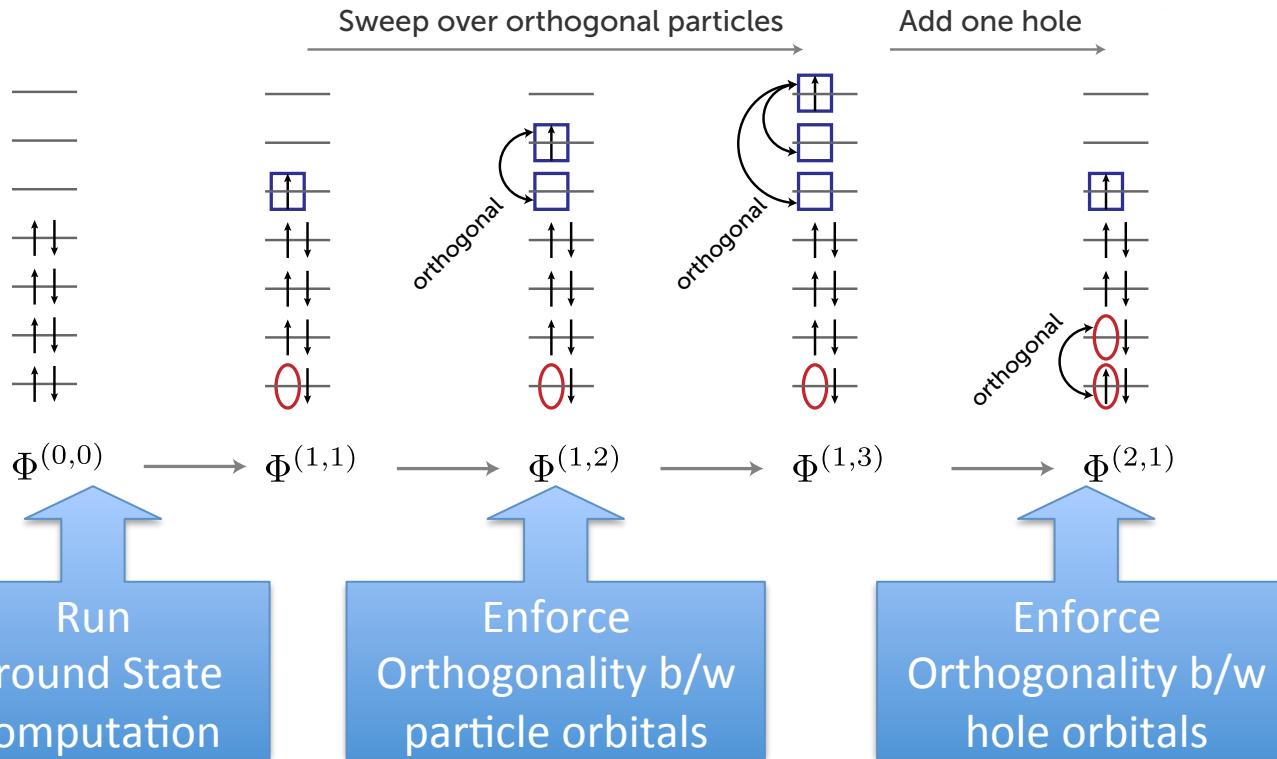
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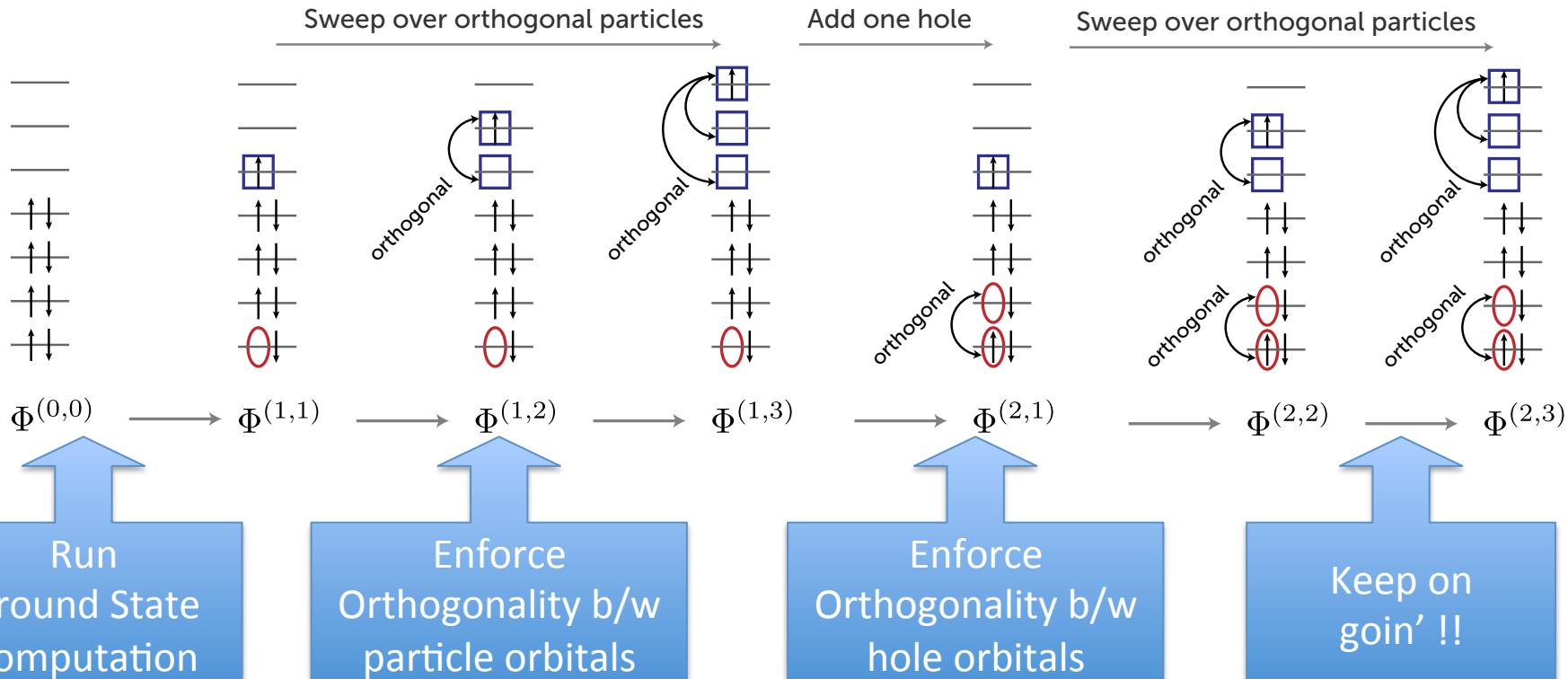
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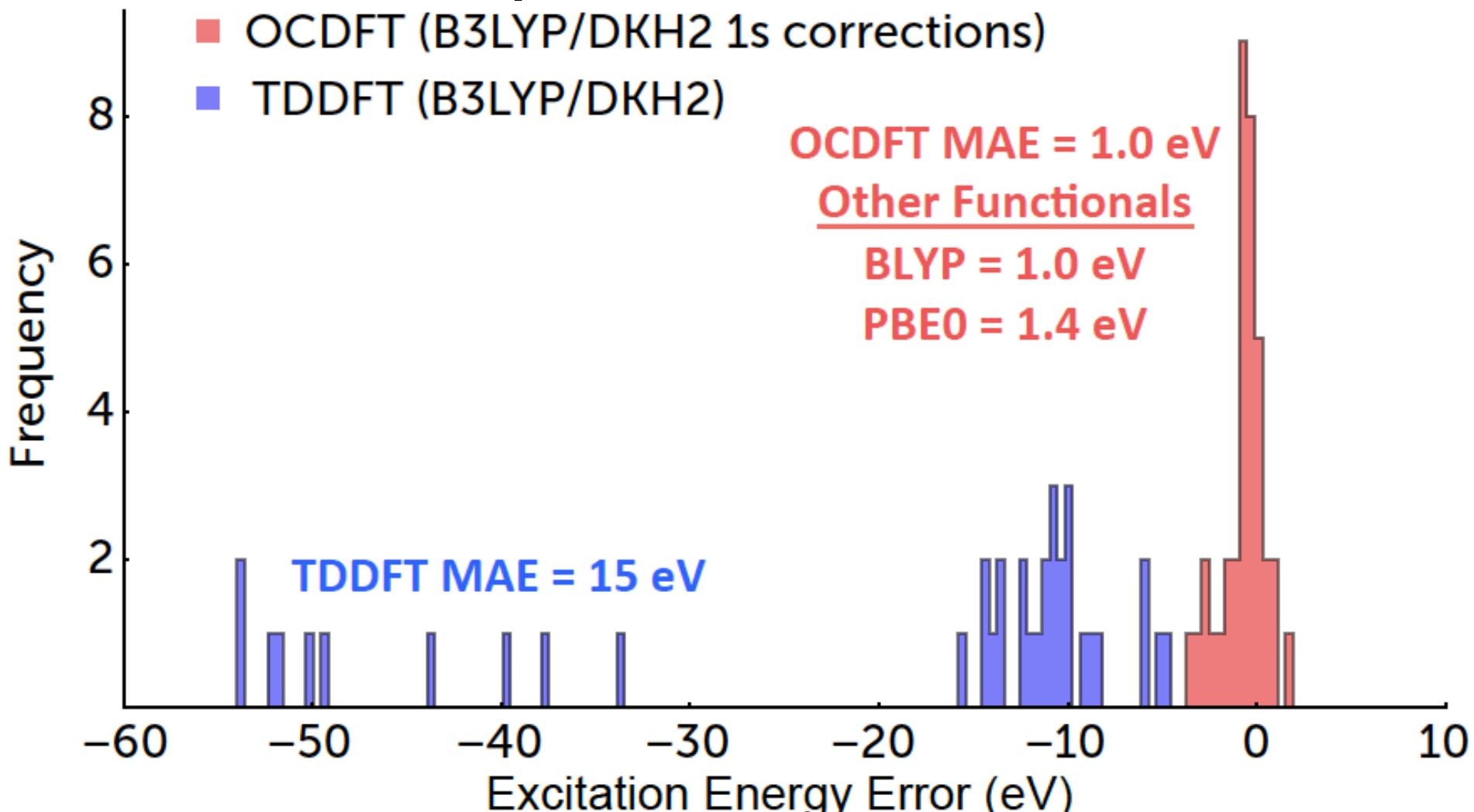


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Comparison to TDDFT

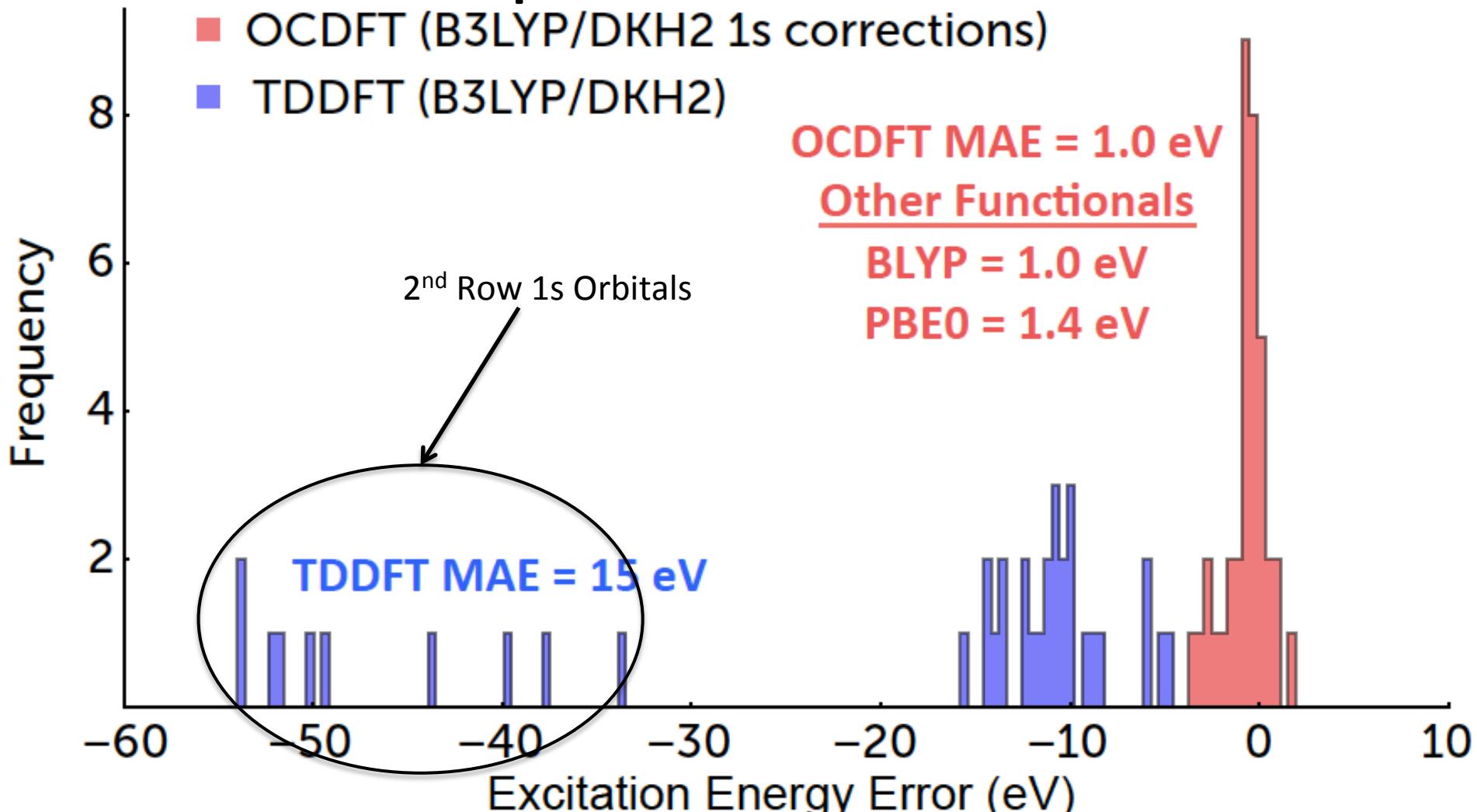


- EOM-CCSD¹ (MAE) = 0.9 eV
- SOS-CIS(D)² [MAE] = 1.2 eV
- Max Error for OCDFT: -3.7 eV
- Max Error for TDDFT: -53.6 eV

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Sensitivity to Orbital Overlap

- Comparing the sensitivity of OCDFT and TDDFT using Tozer Overlap Metric.¹

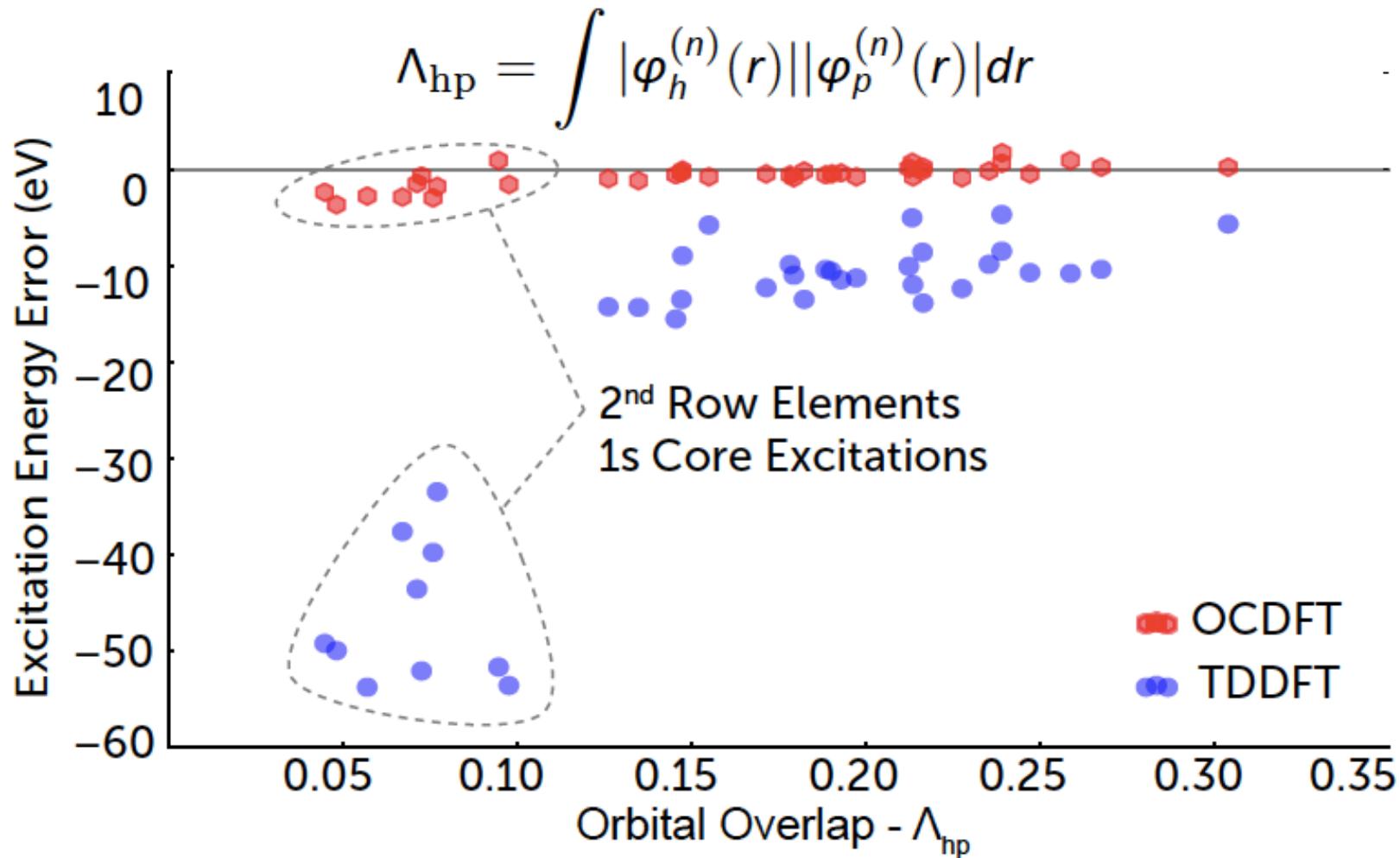
$$\Lambda_{hp} = \int |\varphi_h^{(n)}(r)| |\varphi_p^{(n)}(r)| dr$$

-

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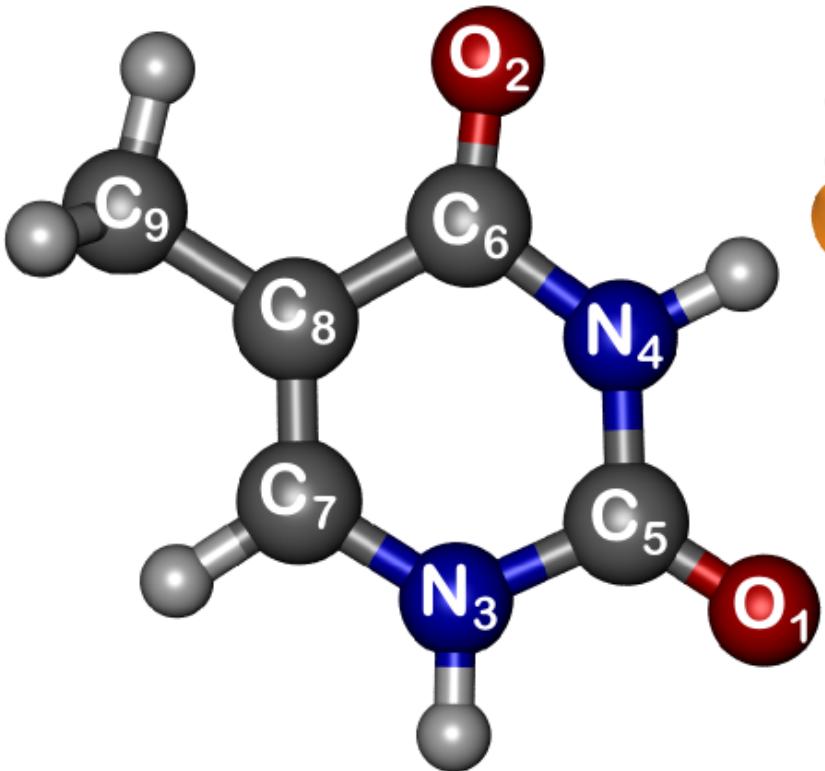
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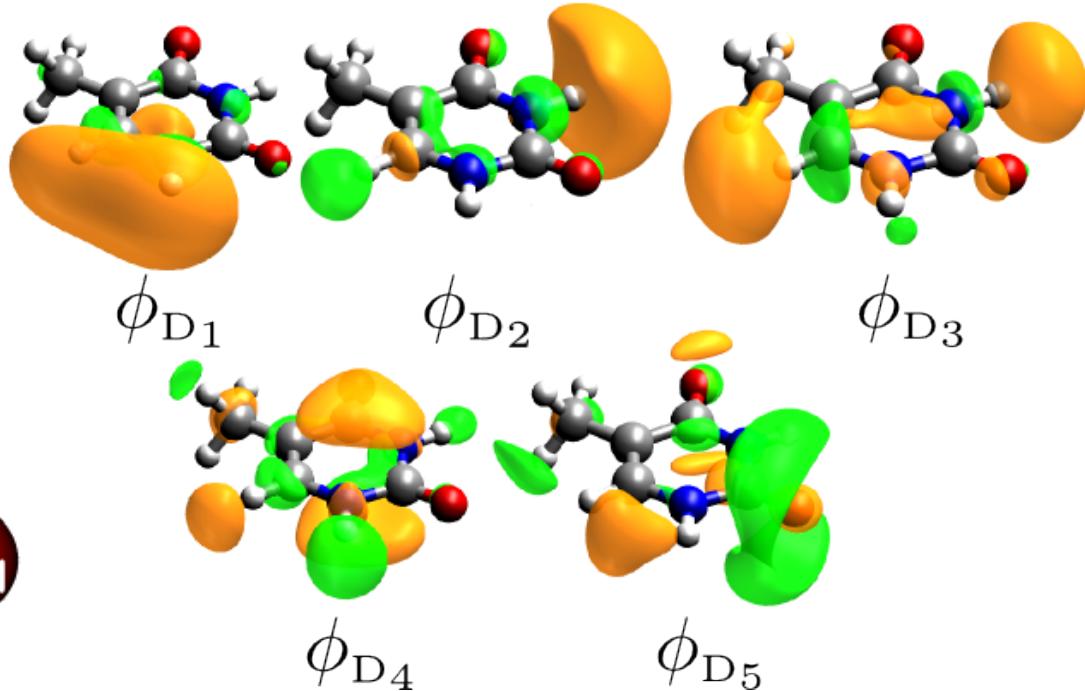


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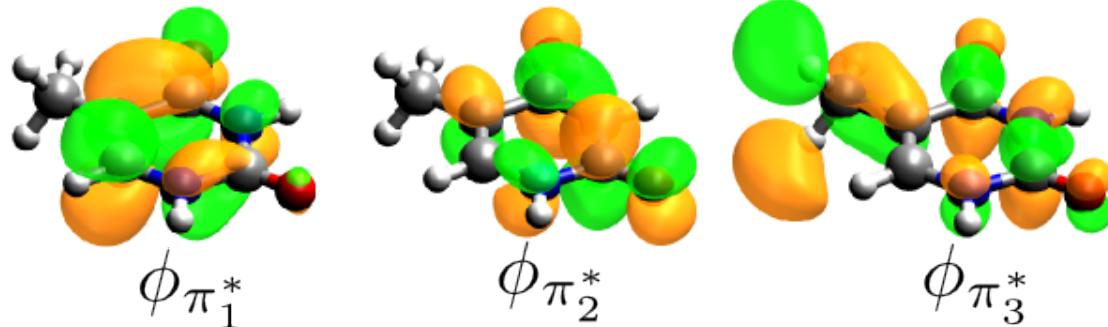
Application to Thymine



Diffuse Orbitals



Anti-Bonding Orbitals



Spectral Simulation Details

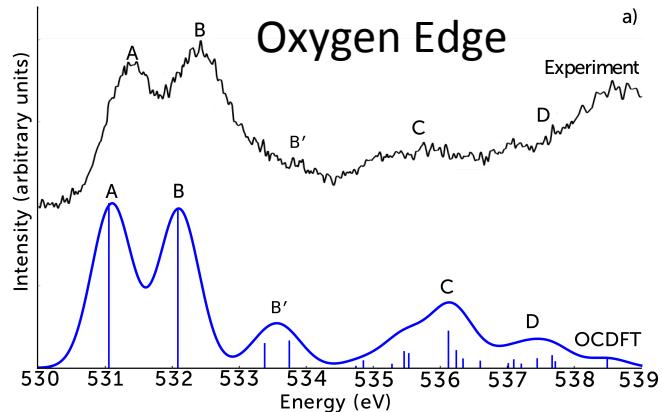
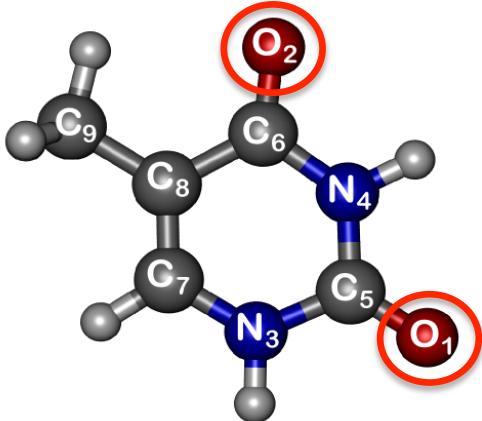
- Compute 10 excitations per hole for each carbon, nitrogen, and oxygen 1s orbital in thymine.
- Plot spectra using gaussians with FWHM of 0.1 eV - 0.4 eV in order to simulate natural spectroscopic broadening effects.
- Compare the OCDFT results with experiment and previously applied 2nd order algebraic diagrammatic construction [ADC(2)] methods
- Transition dipole moments are approximated using Kohn-Sham determinants and the position vector

$$\mu_{fi} = \langle \Phi^{(f)} | \hat{r} | \Phi^{(i)} \rangle$$

- Using the approximate transition dipole moment we can now calculate an oscillator strength for each transition

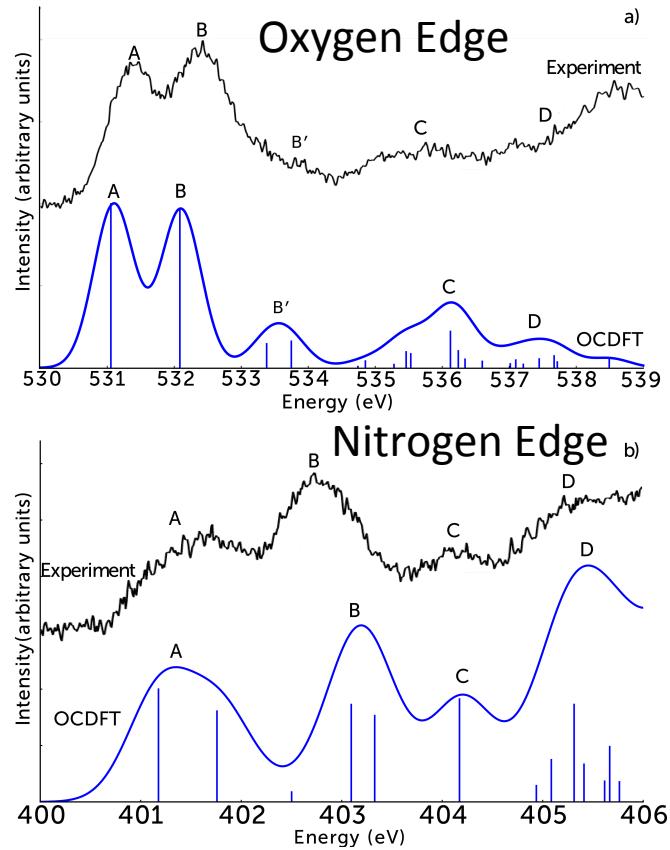
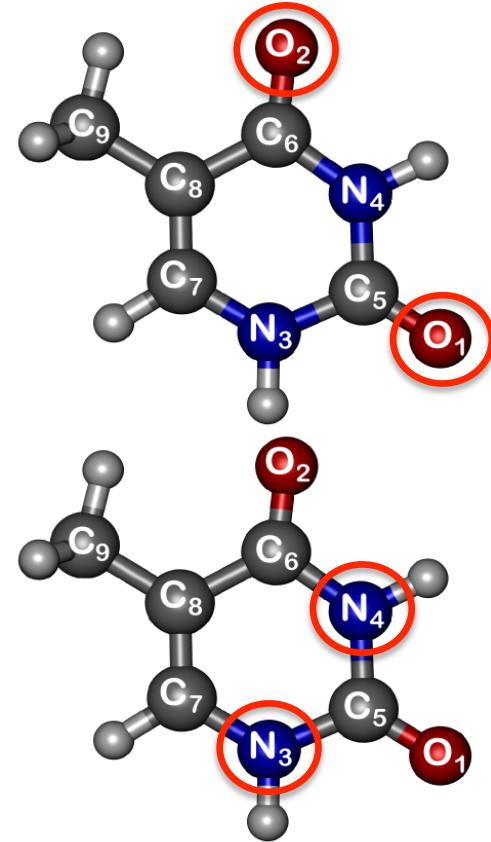
$$f_{\text{osc}} = \frac{2}{3} |\mu_{fi}|^2 \omega_{fi}$$

OCDFT Simulation of Thymine K-Edge



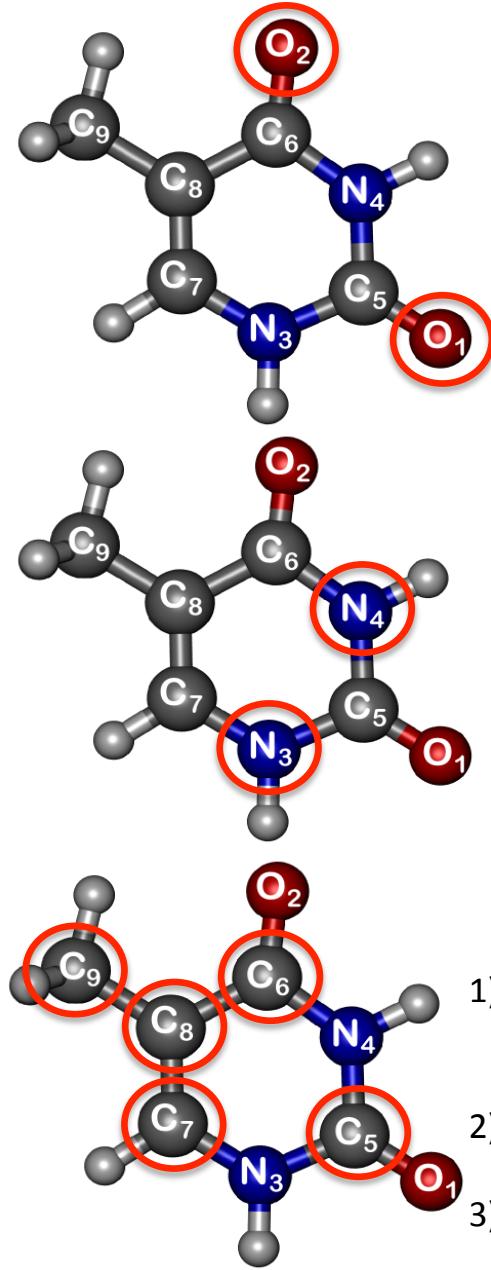
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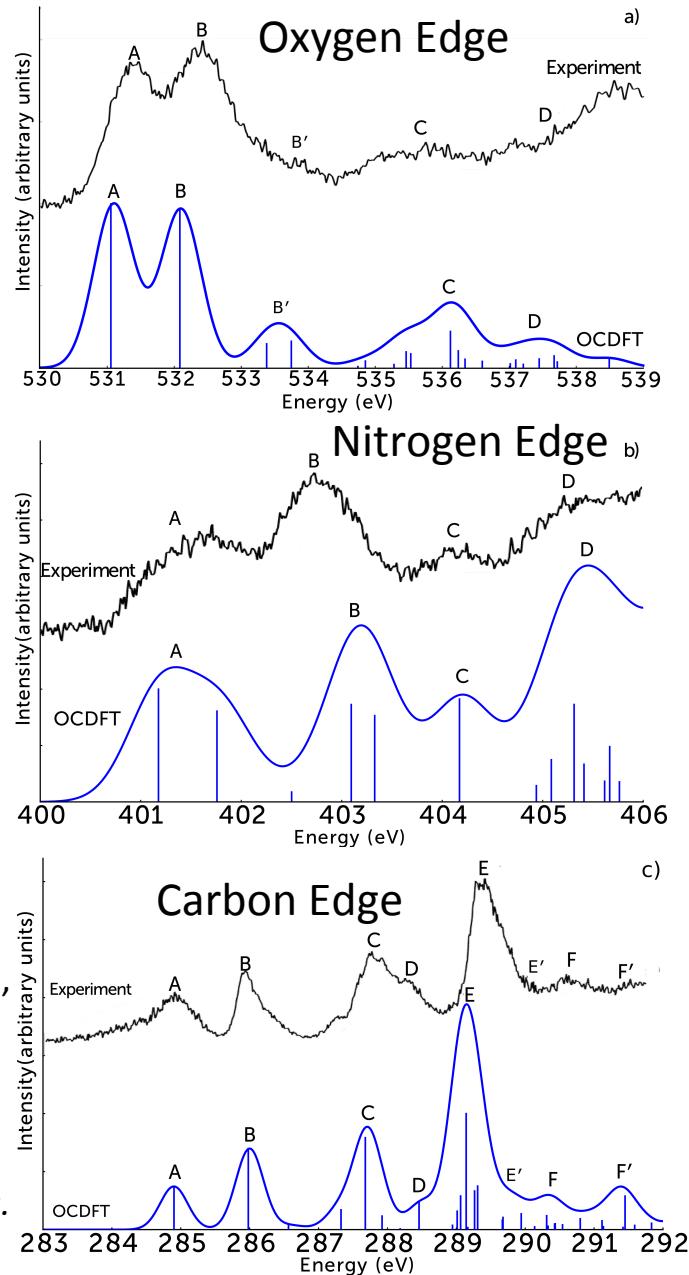


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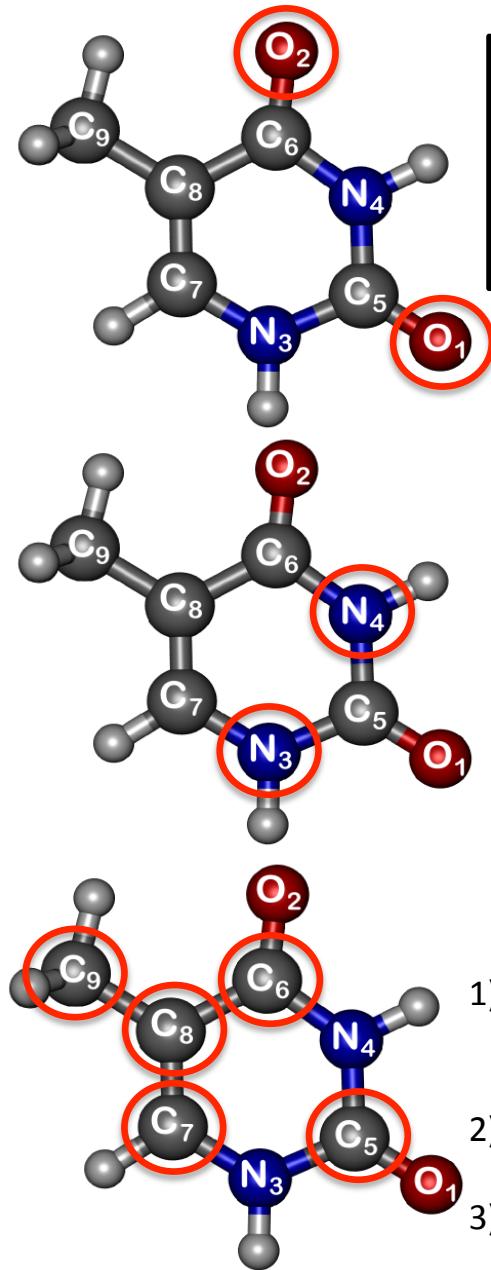
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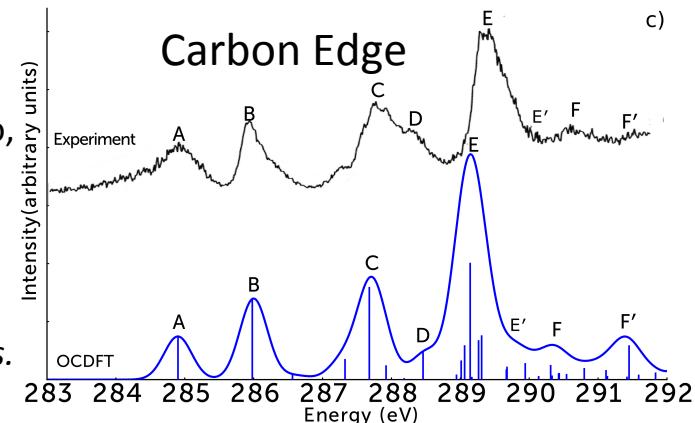
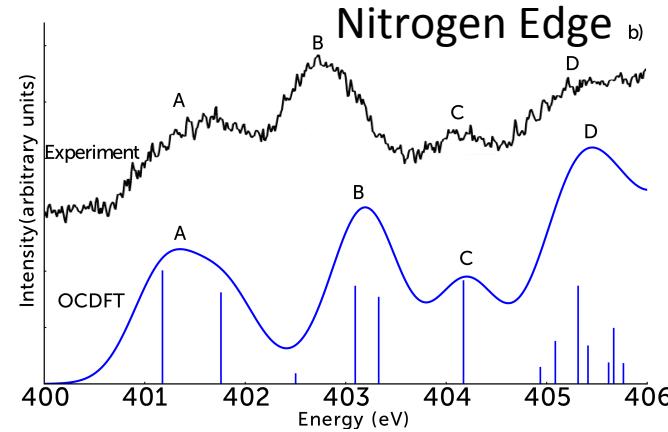
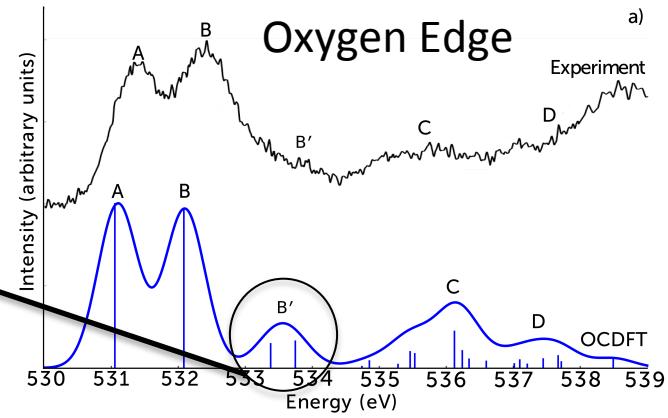
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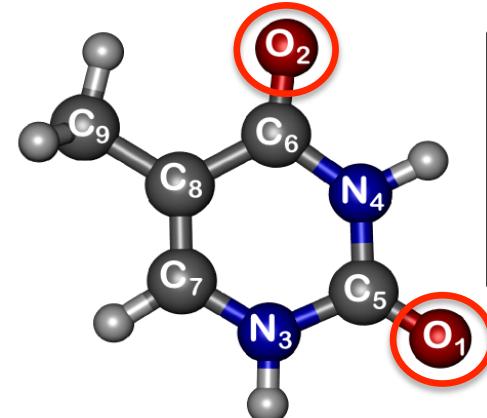


Shoulder Feature B' initially absent in ADC(2) spectrum.¹
Was confirmed by later CVS-ADC(2) study.²

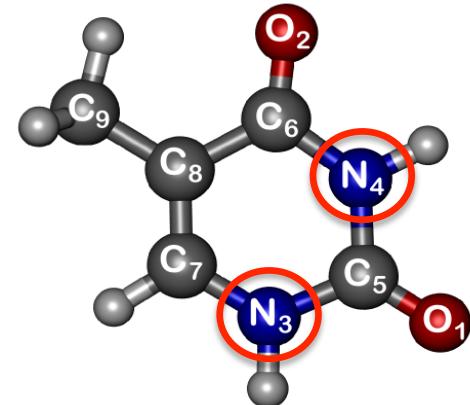


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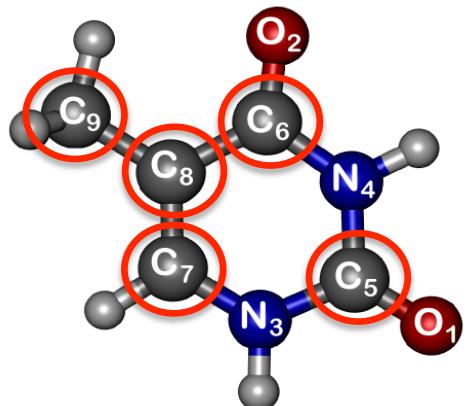
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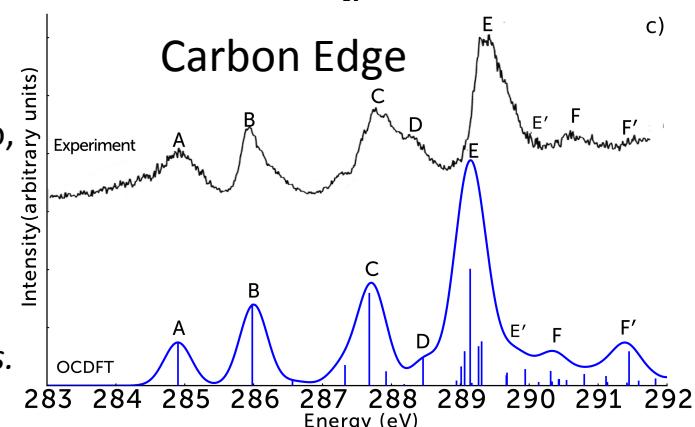
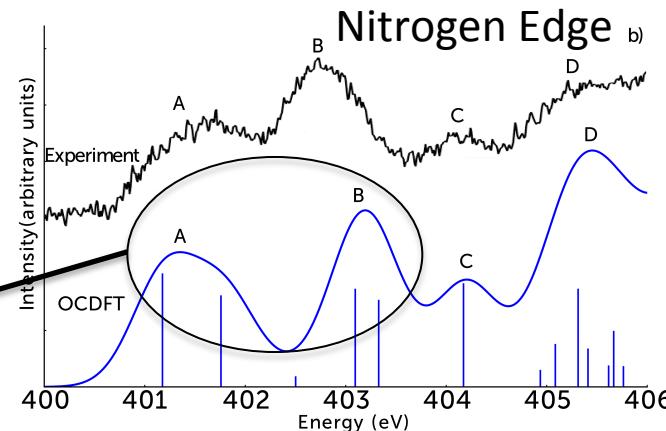
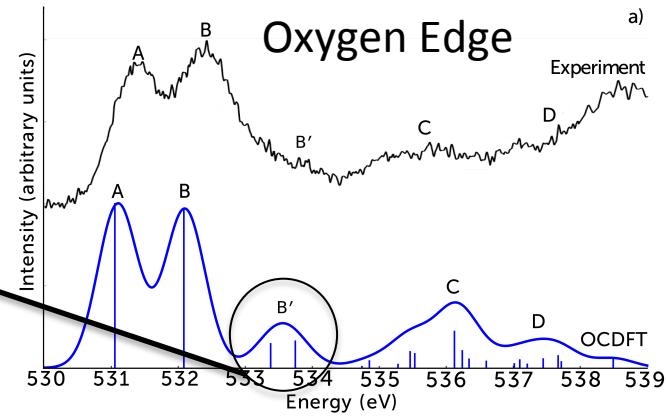
Shoulder Feature B' initially absent in ADC(2) spectrum.¹
Was confirmed by later CVS-ADC(2) study.²



Peak Features A and B are unresolved in ADC(2) spectrum. Appear as single peak feature.

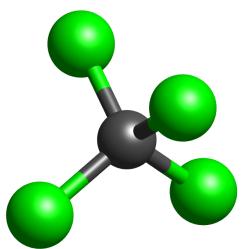


- 1) Plekan, O.; Feyer, V.; Richter, R.; Coreno, M.; de Simone, M.; et al. *Chem. Phys.* 2008
- 2) Wenzel, J.; Wormit, M.; Dreuw, A. J. *Comput. Chem.* 2014
- 3) Derricotte, W. D.; Evangelista, F. A. *Phys. Chem. Chem. Phys.* 2015

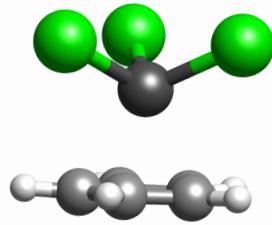


Application to Transition Metal Complexes

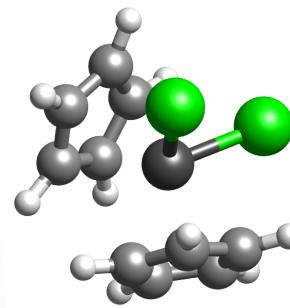
Ti K-Edge of Tetra coordinated Titanium Complexes



TiCl_4



TiCpCl_3

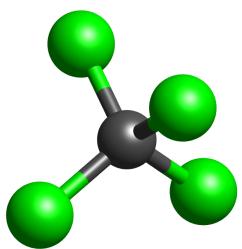


TiCp_2Cl_2

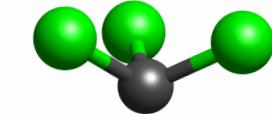
- Class of molecules used to study covalency in cyclopentadienyl (Cp) complexes relevant for use as anti-cancer drug.
- Full treatment of scalar relativistic effects will be handled by X2C Hamiltonian implemented as plugin in PSI4

- 1) DeBeer George, S.; Brant, P.; Solomon, E. I. *J. Am. Chem. Soc.* **2005**
- 2) Casarin, M.; Finetti, P.; Vittadini, A.; Wang, F.; Ziegler, T. *J. Phys. Chem. A* **2007**.

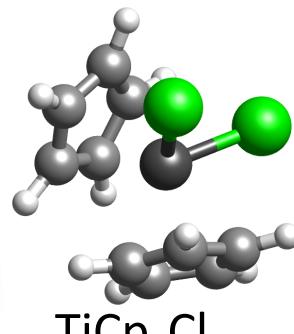
Ti K-Edge of Tetra coordinated Titanium Complexes



TiCl_4

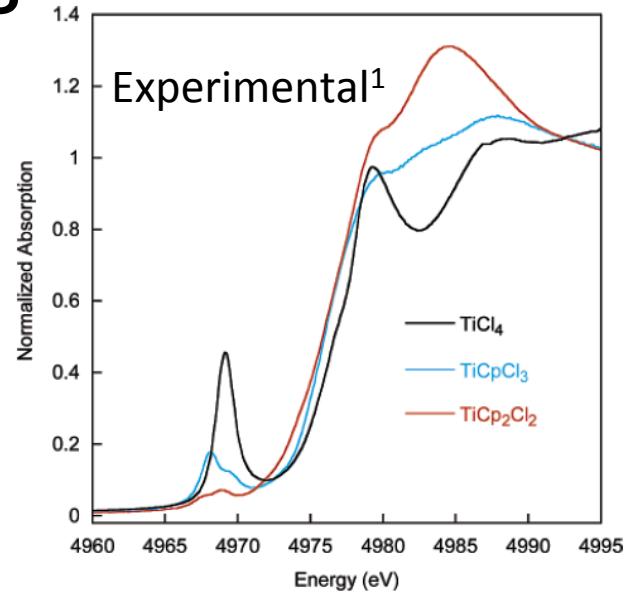


TiCpCl_3



TiCp_2Cl_2

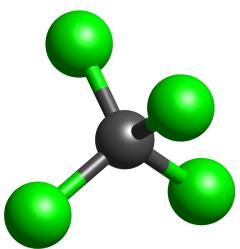
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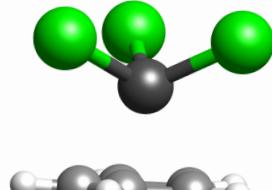
1) DeBeer George, S.; Brant, P.; Solomon, E. I. *J. Am. Chem. Soc.* **2005**

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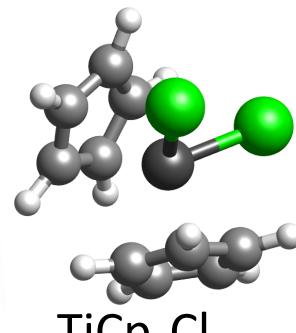
Ti K-Edge of Tetra coordinated Titanium Complexes



TiCl_4

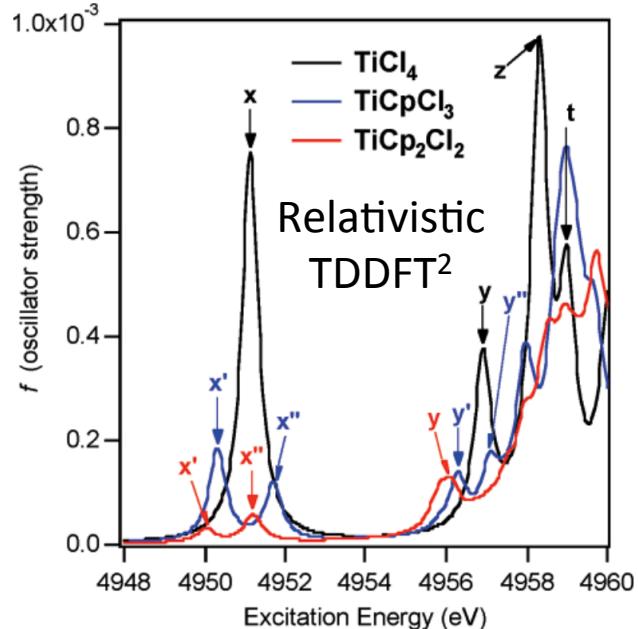
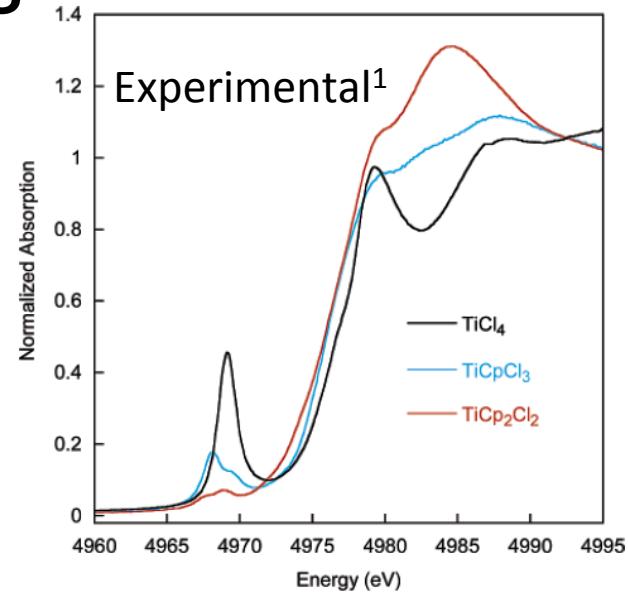


TiCpCl_3



TiCp_2Cl_2

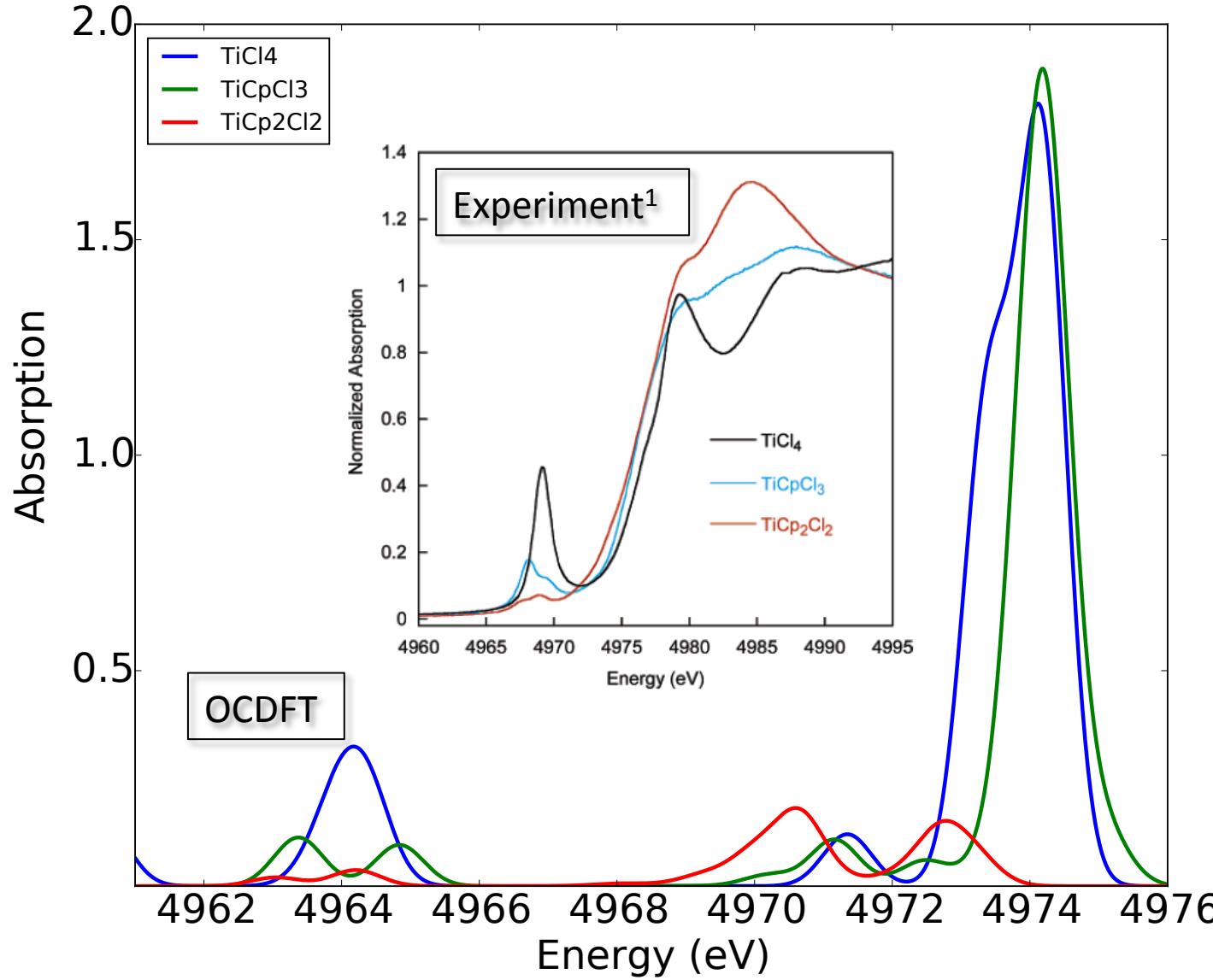
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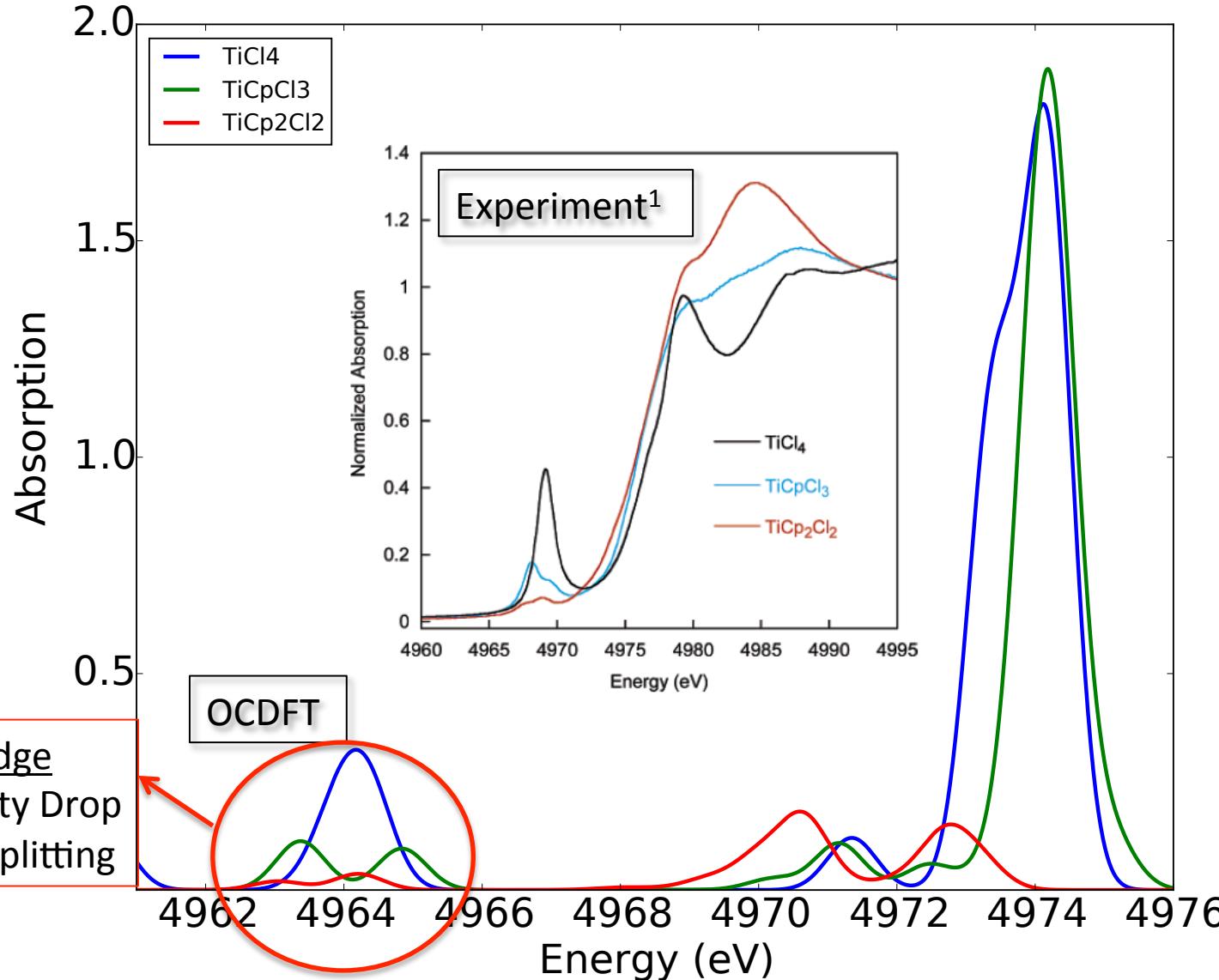
Comparison With Experimental Spectra



- Why are the peaks split for the Cp compounds? Maybe OCDFT Particle Orbitals can shed some insight?

1) DeBeer George, S.; Brant, P.; Solomon, E. I. *J. Am. Chem. Soc.* 2005

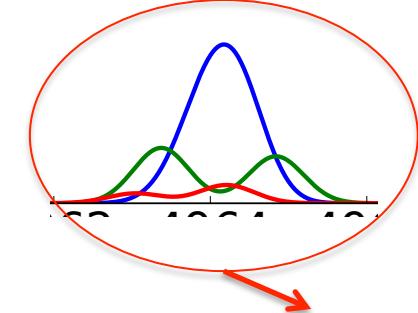
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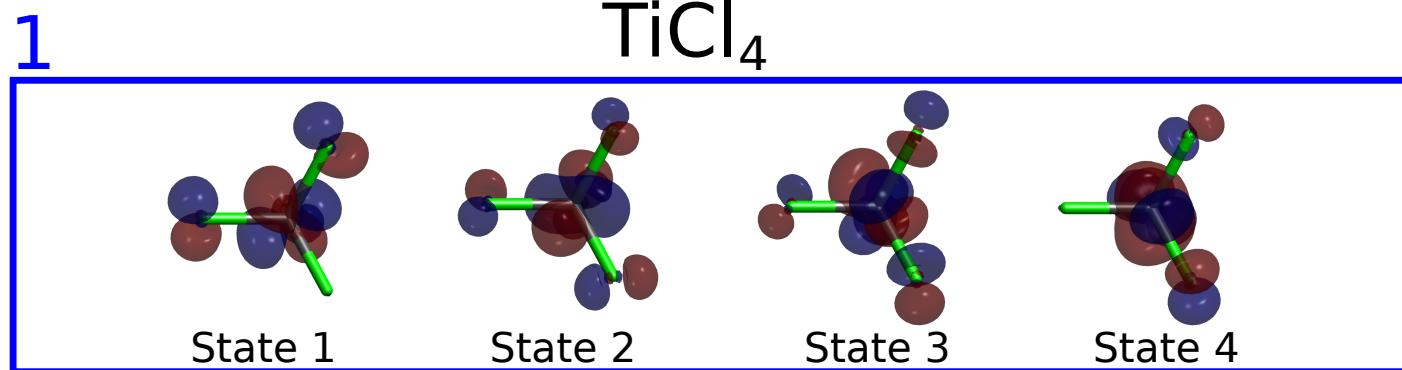
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Pre-Edge Analysis

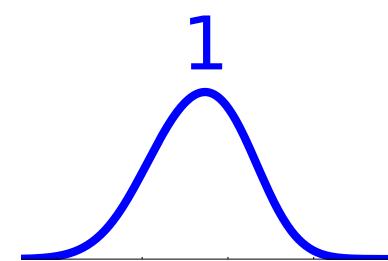


State	Transition Energy (eV)	Rel. Osc. Strength
TiCl_4		
1	4963.85	0.0805
2	4963.53	0.0181
3	4964.38	0.1983
4	4964.03	0.1120

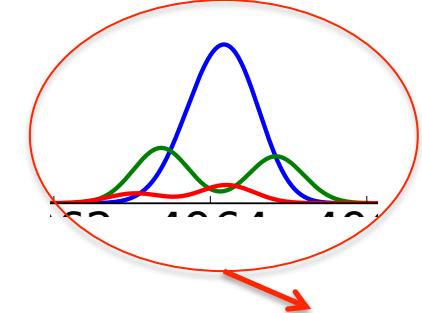
Particle Orbitals



Pre-Edge

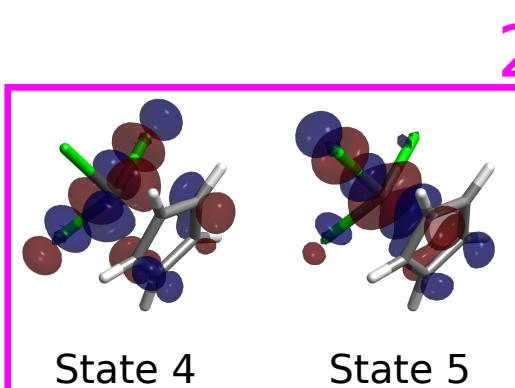
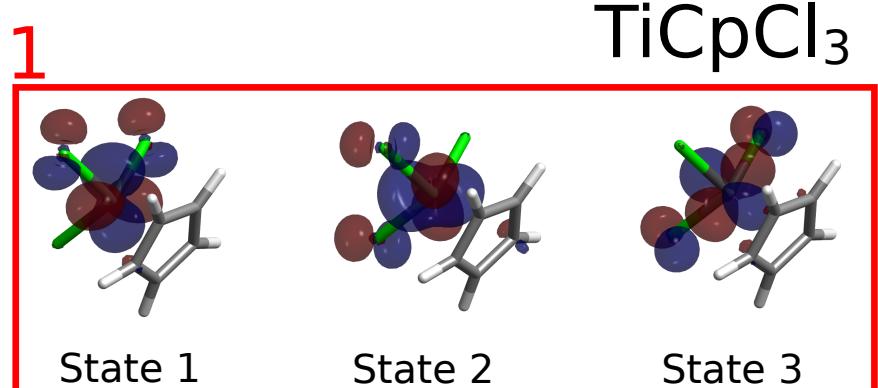


Pre-Edge Analysis



State	Transition Energy (eV)	Rel. Osc. Strength
TiCpCl₃		
1	4963.31	0.0226
2	4963.38	0.0227
3	4963.45	0.0190
4	4964.80	0.0271
5	4964.88	0.0270

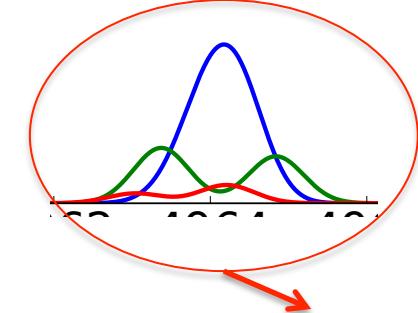
Particle Orbitals



Pre-Edge

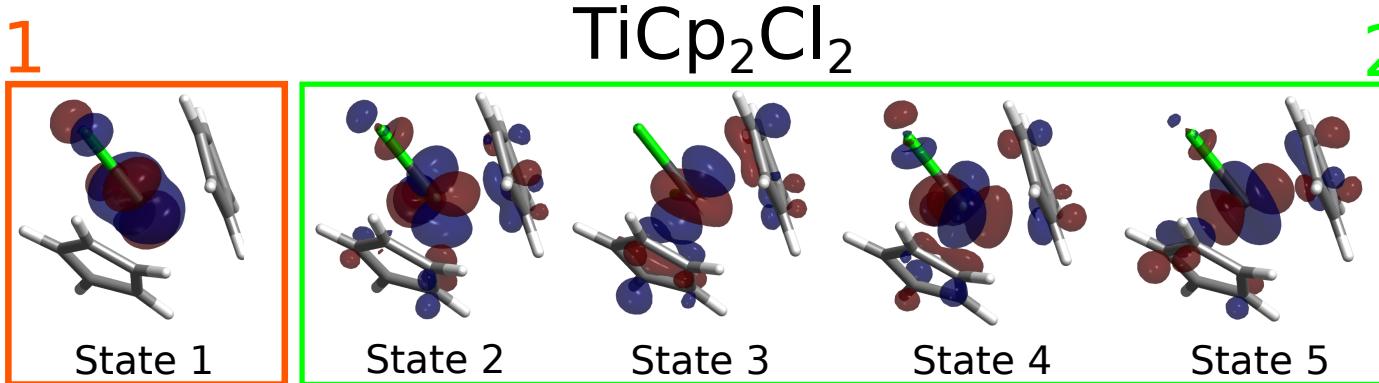


Pre-Edge Analysis

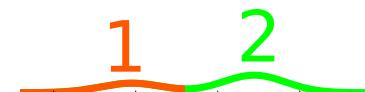


State	Transition Energy (eV)	Rel. Osc. Strength
TiCp₂Cl₂		
1	4963.04	0.0114
2	4964.20	0.0079
3	4964.18	0.0012
4	4964.32	0.0003
5	4964.22	0.0116

Particle Orbitals



Pre-Edge



Conclusions

- Extended OCDFT to calculate multiple excited states in order to fully simulate NEXAS spectra.
- OCDFT has no significant dependence on the amount of Hartree-Fock exchange present in the functional
- Looked at excitations from first and second row elements and show that OCDFT is less sensitive to changes in orbital overlap.
- Calculated the NEXAS spectra of thymine in order to show that OCDFT is a useful tool for interpreting NEXAS spectra.
- Coupled OCDFT with X2C relativistic Hamiltonian in order to show that OCDFT can be effective at simulating NEXAS spectra of transition metal complexes.

Thank You



Evangelista Lab Members

Francesco Evangelista ← Advisor

Prakash Verma
Chenyang Li

Kevin Hannon
Jeff Schriber
Tianyuan Zhang

← Post-Docs

← Minions



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